

Determination of Available Chlorine in Bleaching Powder

Principle

Bleaching powder is commonly used as a disinfectant. The chlorine present in the bleaching powder gets reduced with time. So, to find the exact quantity of bleaching powder required, the amount of available chlorine in the sample must be found out.

Chlorine will liberate free iodine from potassium iodide solution when its pH is 8 or less. The iodine liberated, which is equivalent to the amount of active chlorine, is titrated with standard sodium thiosulphate solution using starch as indicator.

Bleaching powder

Bleaching powder is also referred to as T.C.L. powder (**Terephthaloyl chloride**). Bleaching powder is nothing but a compound of Lime and Chlorine.

In fresh Bleaching powder, normally chlorine content is 33 % or more. Bleaching powder is unstable in nature and as the time passes the chlorine content reduces. To avoid loss of chlorine, it is necessary to store Bleaching powder in dry condition and away from direct sun light. If the Bleaching powder is in a bag, after use, the bag should be properly closed.

- Contact period of minimum half an hour is required for the chlorine to act on the pathogens and kill them. Hence, it is necessary to do the chlorination, minimum one hour before the supply hours.
- The quantity of Bleaching powder required for disinfection depends on the quantity of water in the GSR/ESR. It can be found out with the help of a formula.

Chlorination:

Quantity of Bleaching Powder required for disinfection as per above formula should be taken in a plastic bucket and adding small quantity of water, a thick paste should be prepared. Then the bucket should be filled up to 50% to 75% by water and the paste should be thoroughly

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