

Data And Lab Submission - Determination Of Solubility

Product Constant

Determination of a Solubility Product Constant

Are you completing this experiment online?



Yes

Data Collection

Concentration of standard HCl solution (M)

0.050

Volume and temperature measurements

	Trial 1	Trial 2
Initial burette reading (mL)	1.16	1.88
Final burette reading (mL)	12.21	13.12
Solution temperature (°C)	25.0	25.0

(9pts) Calculations

Use the data collected above to complete the following calculations.

Determination Of A Solubility Product Lab Answers

Wolfgang Guggemos



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