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atomic number 9 mass number 18 charge of 1 b atomic number 43 mass number 99 charge of 7 c atomic number 53 atomic

mass number 131 charge of 1

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mass isotope mass defect mass number mass spectrometer nuclear binding energy unified atomic mass unit weighted

average

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web oct 23 2021 1 what particles determine the mass number 2 why is mass number always a whole number 3 one isotope of carbon c has exactly the same mass number and atomic mass sin used as the definition of the atomic mass unit amu which isotope is it and what atomic mass 4 what is the approximate mass of one proton amu 5

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web define isotope using mass number atomic number number of protons neutrons and electrons given information about an element find the mass and name of an isotope give evidence to support or dispute in nature the chance of finding one isotope of an element is the same for all isotopes

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web the atomic mass of a single atom is simply its total mass and is typically expressed in atomic mass units or amu by definition an atom of carbon with six neutrons carbon 12 has an atomic mass of 12 amu other atoms don t generally have round number atomic masses for reasons that are a little beyond the scope of this article

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web jul 29 2021 as you learned the mass number is the sum of the numbers of protons and neutrons present in the nucleus of an atom the mass number is an integer that is approximately equal to the numerical value of the atomic mass although the mass number is unitless it is assigned units called atomic mass units amu because a molecule or a

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