

- 2-1. Water at 120 C and a pressure of 250 kPa passes through a pressure-reducing valve and then flows to a separating tank at standard atmospheric pressure of 101.3 kPa, as shown in Fig. 2-14.
- (a) What is the state of the water entering the valve (subcooled liquid, saturated liquid, or vapor)?
- (b) For each kilogram that enters the pressure-reducing valve, how much leaves the separating tank as vapor?



Fig. 2-14. Pressure-reducing valve in Problem 2-1

Solution:

- (a) From Fig. 2-2, a temperature of 120 C and pressure of 250 kPa water lies in the sub-cooled region, so it is a sub-cooled liquid.
- (b) At 120 C, $h_1 = 503.72$ kJ/kg from Table A-1

For Pressuring Reducing Valve $Dh = 0$
 $h_2 = h_1$

At 101.3 kPa, Table A-1, $h_g = 419.06$ kJ/kg
 $h_f = 2676$ kJ/kg

Let x be the amount of vapor leaving the separating tank.

$$h = h_f + x(h_g - h_f)$$

$$x = \frac{h - h_f}{h_g - h_f} = \frac{503.72 - 419.06}{2676 - 419.06}$$

$$x = 0.0375 \text{ kg/kg} \dots \text{Ans.}$$

- 2-2. Air flowing at a rate of 2.5 kg/s is heated in a heat exchanger from -10 to 30 C. What is the rate of heat transfer?

Solution:

$$q = mc_p(t_2 - t_1)$$

$m = 2.5$ kg/s
 $c_p = 1.0$ kJ/kg.K
 $t_2 = 30$ C
 $t_1 = -10$ C

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