

Quiz di Chimica

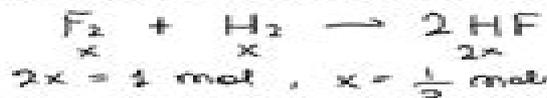
1. La fusione del ghiaccio è una reazione: -0.12

- a) Isoentalpica
- b) Esotermica
- c) Nessuna delle precedenti
- d) IsoEntropica

ENDOTERMICA

2. Quanto fluoro gassoso deve reagire con idrogeno gassoso per dare esattamente una mole di HF?

- a) 1 mole
- b) 2 moli
- c) 1 litro
- d) Nessuna delle altre



3. La configurazione elettronica: $[\text{Ar}]3d64s0$ rappresenta lo ione:

- a) Mn^{2+}
- b) Ni^{2+}
- c) Fe^{3+}
- d) Co^{3+}
- e) Cu^{2+}



4. Quanti legami covalenti forma Xe per completare l'ottetto?

- a) 0
- b) 1
- c) 2
- d) 3
- e) 4

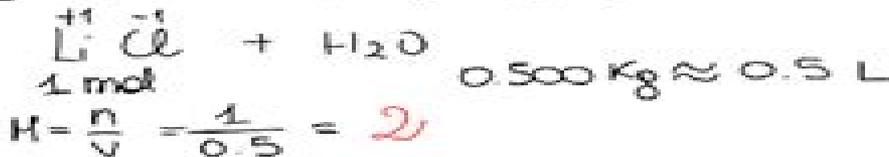
5. Pressione e temperatura in condizioni standard

- a) 1 bar e 25°C
- b) 1 atm e 0°C
- c) 1 atm e 25°C



6. Una mole di cloruro di litio viene disciolta in 0,500 Kg di acqua. Qual è la concentrazione della soluzione risultante?

- a) 0,5 mol/l
- b) 0,5 m
- c) 1 m
- d) 2 M
- e) 1,5 m



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JS Bruner



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