

Nuclear Chemistry Worksheet 3

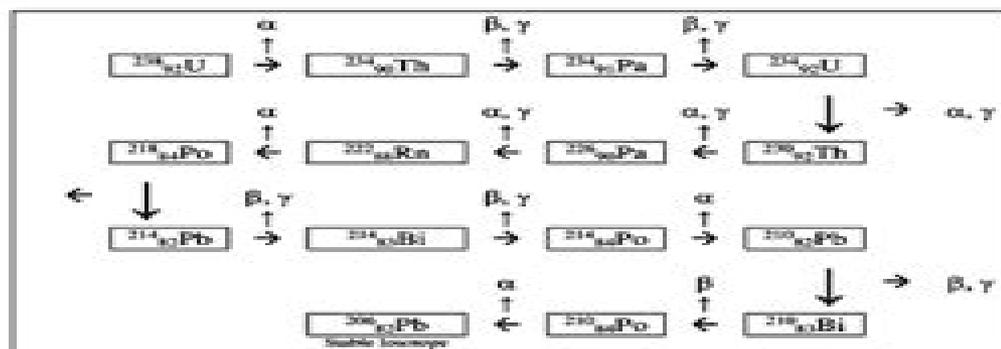
Directions: Identify the following as alpha, beta, gamma, or neutron.

- | | | | |
|-------------------------|--------------------------|--------------------------|------------------------------|
| 1. $\frac{1}{0}n$ _____ | 2. $\frac{0}{-1}e$ _____ | 3. $\frac{4}{2}He$ _____ | 4. $\frac{0}{0}\gamma$ _____ |
|-------------------------|--------------------------|--------------------------|------------------------------|
-
- | | |
|--|-------|
| 5. Nuclear decay with no mass and no charge | _____ |
| 6. An electron | _____ |
| 7. Least penetrating nuclear decay | _____ |
| 8. Most damaging nuclear decay to the human body | _____ |
| 9. Nuclear decay that can be stopped by skin or paper. | _____ |
| 10. Nuclear decay that can be stopped by aluminum. | _____ |

Complete the following nuclear equations.

- | | |
|--|---|
| 11. ${}_{19}^{42}K \rightarrow {}_{-1}^0e +$ _____ | 12. ${}_{94}^{239}Pu \rightarrow \frac{4}{2}He +$ _____ |
| 13. ${}_{4}^9Be \rightarrow$ _____ + ${}_{4}^9Be$ | 14. ${}_{92}^{235}U \rightarrow$ _____ + ${}_{90}^{231}Th$ |
| 15. ${}_{3}^6Li \rightarrow$ _____ + $\frac{4}{2}He$ | 16. _____ \rightarrow ${}_{56}^{142}Ba +$ ${}_{36}^{91}Kr + 3 \frac{1}{0}n$ |

Nuclear Decay Series



The figure maps the radioactive decay of uranium-238 to lead-206. Use the figure to answer the following questions.

17. How many alpha particles are produced as one atom of uranium-238 decays to an atom of lead-206? _____
18. How many beta particles? _____
19. Explain why lead-206 is a stable isotope. _____

25 3 Nuclear Chemistry Review Answers Key

Marco Cascella



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