

**UNIT 10: PERIODIC TABLE** **PERIODIC TABLE** **PERIODIC TABLE** **PERIODIC TABLE** **PERIODIC TABLE**

**Section 10.1** *(continued)*

In your notebook, read aloud the modern periodic table.

Copy the letterheads in the boxes on the left below from the periodic table to complete the table on the right.

_____
_____
_____
_____
_____
_____

atomic mass	_____
atomic number	_____
electron configuration	_____
chemical name	_____
chemical symbol	_____

Put each item in Column A, under the letter of the matching item in Column B.

**Column A**

**Column B**

- |   |                                  |
|---|----------------------------------|
| _____ 18. A column in the periodic table                  | _____ a. metals                  |
| _____ 19. A row in the periodic table                     | _____ b. group                   |
| _____ 20. Group 1 elements                                | _____ c. period                  |
| _____ 21. Elements that are shiny and conduct electricity | _____ d. representative elements |
| _____ 22. Group 8 elements                                | _____ e. transition elements     |

In the space on the left, write each of the statements in terms of the statement in bold. Change the underlined word or phrase to make it true.

- \_\_\_\_\_ 23. There are two representative elements of platinum.
- \_\_\_\_\_ 24. Most elements besides boron of the elements in the periodic table are metalloids.
- \_\_\_\_\_ 25. Group 1A elements except for hydrogen are located on the left side.
- \_\_\_\_\_ 26. Group 2A elements are the alkaline earth metals.
- \_\_\_\_\_ 27. Group 7A elements are highly reactive nonmetalic elements in halogens.
- \_\_\_\_\_ 28. Group 8A elements are inert nonmetallic elements known as noble gases.
- \_\_\_\_\_ 29. Metalloids have properties of both metals and nonmetals.

# Study Guide For Content Mastery Solutions Answers

**Yi-Tong Ma**



**Study Guide For Content Mastery Solutions Answers:**

## Reviewing **Study Guide For Content Mastery Solutions Answers**: Unlocking the Spellbinding Force of Linguistics

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2  $3\text{Ca(OH)}_2 + 1\text{Al}_2\text{SO}_4 \rightarrow 3\text{CaSO}_4 + 2\text{Al(OH)}_3$  type of reaction double displacement  
3  $3\text{Mg} + 1\text{Fe}_2\text{O}_3 \rightarrow 2\text{Fe} + 3\text{MgO}$  type of reaction single displacement  
4  $1\text{C}_2\text{H}_4 + 3\text{O}_2 \rightarrow 2\text{CO}_2$

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1  $1\text{CaF}_2 + 2\text{Na} \rightarrow 2\text{NaF} + 1\text{Ca}$  type of chemical reaction single displacement  
2  $1\text{CaCO}_3 \rightarrow 1\text{CaO} + 1\text{CO}_2$  type of chemical reaction decomposition

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