



Fig 5.3. Energy Levels

- The first energy level (labelled 1) can hold up to only two electrons.
- The second energy level (labelled 2) can hold a maximum of eight electrons. This energy level is filled after the first energy level and before the third level.
- The third energy level (labelled 3) can hold a maximum of 18 electrons, however when 8 electrons are in the third level there is a degree of stability; and other electrons are added to the fourth energy level.
- Electrons must first fill the first energy level before they start occupying the second energy level.

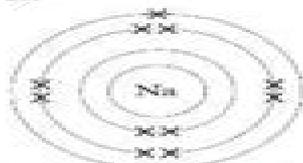


Fig 5.4 Shell model for sodium (Na)

Number of protons = 11, number of electrons = 11

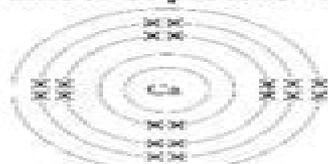


Fig 5.5 Shell model for calcium (Ca)

Number of protons = 20, number of electrons = 20

Atomic number, mass numbers, sub-atomic particles and electronic configuration of the first 20 elements in the Periodic Table

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Table of Contents Olevel Intergrated Science Notes

1. Understanding the eBook Olevel Intergrated Science Notes
 - The Rise of Digital Reading Olevel Intergrated Science Notes
 - Advantages of eBooks Over Traditional Books
2. Identifying Olevel Intergrated Science Notes
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Olevel Intergrated Science Notes
 - User-Friendly Interface
4. Exploring eBook Recommendations from Olevel Intergrated Science Notes
 - Personalized Recommendations
 - Olevel Intergrated Science Notes User Reviews and Ratings
 - Olevel Intergrated Science Notes and Bestseller Lists

5. Accessing Olevel Intergrated Science Notes Free and Paid eBooks
 - Olevel Intergrated Science Notes Public Domain eBooks
 - Olevel Intergrated Science Notes eBook Subscription Services
 - Olevel Intergrated Science Notes Budget-Friendly Options
6. Navigating Olevel Intergrated Science Notes eBook Formats
 - ePub, PDF, MOBI, and More
 - Olevel Intergrated Science Notes Compatibility with Devices
 - Olevel Intergrated Science Notes Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Olevel Intergrated Science Notes
 - Highlighting and Note-Taking Olevel Intergrated Science Notes
 - Interactive Elements Olevel Intergrated Science Notes
8. Staying Engaged with Olevel Intergrated Science Notes
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Olevel Intergrated Science Notes
9. Balancing eBooks and Physical Books Olevel Intergrated Science Notes
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Olevel Intergrated Science Notes
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Olevel Intergrated Science Notes
 - Setting Reading Goals Olevel Intergrated Science Notes
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Olevel Intergrated Science Notes
 - Fact-Checking eBook Content of Olevel Intergrated Science Notes
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
- Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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