

## Elements, compounds and reactions test

See how well you have understood the ideas from the chapter.

1. What is a compound? (1)
  - a) Something that is only made up of one type of atom
  - b) **Something that is made up of two or more types of atoms chemically joined together.**
  - c) Something that is made up of two or more elements or compounds that are not chemically joined together.
2. What does the formula of water,  $H_2O$ , tell us about each molecule? (1)
  - a) It has one atom of hydrogen and one of oxygen
  - b) **It has two atoms of hydrogen and one of oxygen**
  - c) It has one atom of hydrogen and two of oxygen
  - d) It has two atoms of carbon and two of oxygen
3. What is a metalloid? (1)
  - a) A cheap metal
  - b) A recently discovered metal
  - c) **An element with some properties of a metal and some of a non-metal**
  - d) An alloy
4. Each element has a unique number, called its... This number increases from left to right across each period. This is an important pattern for the periodic table (1)
  - a) Mass number
  - b) **Atomic number**
  - c) Period number
  - d) Group number
5. List two properties of metals and two for non-metals (4)

**Metals are ductile, conduct electricity, conduct heat, malleable, sonorous, high melting points, shiny. Non-metals are neither strong or shiny, dull, mainly unreactive gases at room temperature, low density, non-ductile, non-malleable, poor conductors**
6. Explain, using examples, the differences between a chemical change and a physical change. (4)

**Chemical change is irreversible**  
**Reasonable example, such as burning toast, frying egg**  
**Physical change is reversible, often involving a change of state**  
**Reasonable example, such as a change of state**

See how you can apply the ideas in this chapter to new situations

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**Xiang Xie**



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