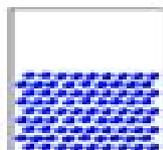


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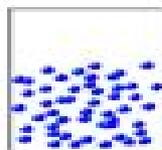
Kinetic Molecular Theory Worksheet

Use the following terms to fill in the blanks. Some may be used more than once. Hopefully all will be included but you never know. If you seem to need an additional term (oops!) just do your best

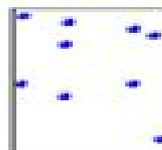
Gas	Solid	Liquid	Phase	Compress
Empty Space	Kinetic Energy	Bonds	Structure	Collisions
Anomaly	Random Motion	Temperature	Absolute Zero	density
Fluids	Compressible	Incompressible		



A



B



C

A	SOLID
B	LIQUID
C	GAS

The boxes above represent matter in the 3 common phases. Write the name of the phase next to the appropriate letter.

- The word kinetic comes from a Greek word that means "to move." The kinetic molecular theory is based upon the assumption that atoms are in constant **Random Motion**.
- State the names of the phases in order of Temperature, Kinetic Energy, density and bond strength between atoms.

Lowest Temperatures	In between Temperatures	Highest Temperatures
Solid	Liquid	Gas
Lowest Kinetic Energy	In between Kinetic Energy	Highest Kinetic Energy
Solid	Liquid	Gas
Lowest Density		Highest Density
Gas		
High Strength Bonds	Weak Bonds	No Bonds
Solid	Liquid	Gas

- Kinetic Energy** is directly related to temperature. As the temperature increases so does the **Kinetic Energy**.
- Compared to Solids and Liquids gases tend to have a lower **density**. This can be explained by the fact that there is so much **Empty Space** between particles in a gas.
- A small pressure will easily **compress** a gas. If this happens its **density** will increase. The space between the atoms will (INCREASE or **DECREASE**).
- Even a large pressure will not **compress** a solid because the atoms are so closely packed.
- Solids have strong **Bonds**, liquids have weaker **Bonds** and gases have no **Bonds** holding them together.
- A solid like ice or diamond has a highly ordered arrangement of bonds between atoms which is called its **Structure**. A liquid or gas does not have this.

Kinetic Molecular Theory Worksheet Answers

Lakhmir Singh & Manjit Kaur



Kinetic Molecular Theory Worksheet Answers:

A Guide to Teaching in the Active Learning Classroom Paul Baepler, J. D. Walker, D. Christopher Brooks, Kem Saichaie, Christina I. Petersen, 2023-07-03 While Active Learning Classrooms or ALCs offer rich new environments for learning they present many new challenges to faculty because among other things they eliminate the room's central focal point and disrupt the conventional seating plan to which faculty and students have become accustomed. The importance of learning how to use these classrooms well and to capitalize on their special features is paramount. The potential they represent can be realized only when they facilitate improved learning outcomes and engage students in the learning process in a manner different from traditional classrooms and lecture halls. This book provides an introduction to ALCs briefly covering their history and then synthesizing the research on these spaces to provide faculty with empirically based practical guidance on how to use these unfamiliar spaces effectively. Among the questions this book addresses are: How can instructors mitigate the apparent lack of a central focal point in the space? What types of learning activities work well in the ALCs and take advantage of the affordances of the room? How can teachers address familiar classroom management challenges in these unfamiliar spaces? If assessment and rapid feedback are critical in active learning, how do they work in a room filled with circular tables and no central focus point? How do instructors balance group learning with the needs of the larger class? How can students be held accountable when many will necessarily have their backs facing the instructor? How can instructors evaluate the effectiveness of their teaching in these spaces? This book is intended for faculty preparing to teach in or already working in this new classroom environment, for administrators planning to create ALCs or experimenting with provisionally designed rooms, and for faculty developers helping teachers transition to using these new spaces. *Tried and True* National Science Teachers Association, 2010. A compilation of popular *Tried and True* columns originally published in *Science Scope*; this new book is filled with teachers' best classroom activities, time-tested, tweaked, and engaging. These ageless activities will fit easily into your middle school curriculum and serve as go-to resources when you need a tried and true lesson for tomorrow. From publisher description: *Lakhmir Singh's Science Chemistry for ICSE Class 8* Lakhmir Singh & Manjit Kaur, Series of books for class 1 to 8 for ICSE schools. The main goal that this series aspires to accomplish is to help students understand difficult scientific concepts in a simple manner and in an easy language. *Holt Chemistry* Ralph Thomas Myers, 2004. **Novice Teacher Action** Anne Liu Kern, 2007. *Merrill Chemistry* Robert C. Smoot, Smoot, Richard G. Smith, Jack Price, 1998. *Educart CBSE Class 9 Science One-shot Question Bank 2026 (Strictly for 2025-26 Exam)* Educart, 2025-06-07. What Do You Get? Question Bank for daily practice. Handpicked important chapter-wise questions. What notable components are included in Educart CBSE CLASS 9 Science ONE SHOT? Chapter-wise concept maps. Each chapter has 3 worksheets for daily practice. Unit-wise worksheets. Pull Out are given separately for extra practice. NCERT Exemplar. DIKSHA PYQs. Competency Based. Important Qs to cover every type of questions. Answer key for every worksheet. Detailed

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