

k _ te

the matching short vowels below

Phonics

A Pull-the-Tab Book



 o x  a g  p a g

Fill in your answers with the wipe-clean marker



Pull the tab, see the answers!

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Reading Randomised Controlled Trials Robert Savage, Amy Fox, Anneka Dawson, Helen Gray, Clare Huxley, 2025-03-03
In early 2021/22 the Flexible Phonics reading intervention a large scale randomised controlled trial took place in 118 schools in England and involved nearly 3 000 children This study aimed to provide valuable insights into the success of largely school based education trials The Flexible Phonics trial proved to be not only an important experiment in improving children s literacy but a case study in which the methodology of single randomised controlled trials in education can be considered Reading Randomised Controlled Trials investigates the complexities of conducting randomised controlled trials in the field of education and how they can be seen as a cultural activity The book emphasises the human operations decision making and actions that drive such trials which are often overlooked in published reports Through the case study of Flexible Phonics the book highlights the key differences between scaled educational trials and other types of trials such as pharmaceutical trials The additional focus on early childhood literacy is of significant educational importance particularly in the context of UK school partial closures in 2020 and 2021 during the Covid 19 pandemic Reading Randomised Controlled Trials is ultimately a unique resource on the implementation of randomised controlled trials in authentic school contexts and for the sustained improvement of practice in education *Houghton Mifflin Reading*, 2004 *Finger Play Reader* John Walter Davis, Fanny Julien, 1909 **Elementary English**, 1955 SCC library has 1949 cur *The Techne*, 1930 **Physical Reconstruction and Vocational Education, Part One** United States. Surgeon-General's Office, Arthur Griswold Crane, 1927 Kindergarten and First Grade, 1919 **Five Messages to Teachers of Primary Reading** Nettie Alice Sawyer, Nettie Sawyer Funk, 1913 **Report of the Minister for Education and Cultural Activities** Queensland. Department of Education, 1927 **Plans and Activities for the Primary Grades** Belle L. Dickson, 1930 **Proceedings and Addresses** American Association on Mental Deficiency, 1930 **The Kindergarten and First Grade**, 1923 **Contributions to Education**, 1928 **Baltimore Bulletin of Education**, 1924 Cosmopolitan Shorthand Writer, 1886 **Reading Process and Practice** Constance Weaver, 2002 Weaver incorporates especially timely information accurate data informed critique and results of often ignored research to help teachers counteract government and corporate intrusion into classrooms **Documents** Boston (Mass.). School Committee, 1916 **New York Teachers' Monographs**, 1921 **The Elementary English Review**, 1929 *Teachers' Monographs*, 1919

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web $m \times c \times \Delta t$ equation 1 where m is the total mass of the solution solute plus solvent c is the specific heat of the solution and Δt is the observed temperature change the specific heat of the solution is generally assumed to be the same as that of

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web the heat of reaction ΔH_{soln} is written after the products in units of kJ mol rxn for example sodium hydroxide dissolves exothermically releasing 44.2 kilojoules per mole dissolved this process is represented as $\text{NaOH (s)} \rightarrow \text{Na}^+ \text{(aq)} + \text{OH}^- \text{(aq)}$ $\Delta H_{\text{soln}} = 44.2 \text{ kJ mol rxn}$ write equations to similarly represent the dissolving process for

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web hand warmers are usually placed inside the gloves to keep your hands warm the heat keeping your hands warm is formed as a result of an exothermic reaction which occurs in a small packet hand warmers have 5 main ingredients in them salt carbon polypropylene vermiculite water and iron

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web background the hand warmers are filled with sodium acetate click the metal disc and you will notice the liquid changes state and becomes a solid in the process heat is released the sodium acetate is a supersaturated solution

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web oct 9 2021 sodium acetate coin or other small piece of metal plastic bags seal the coin inside a small bag place this bag inside a larger bag containing supersaturated sodium acetate get this by dissolving as much sodium acetate as possible in hot water this chemical hand warmer is reusable

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web the goal of my experiment and research is to evaluate the various number of compounds to be used as hand warmers and locate the ones that can raise the temperature to 20 degrees celsius in 100 g water and costs as little as possible

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web in this lab you are challenged to use chemistry to determine which ionic compound makes the best hand warmer in this experiment you will collect data that will allow you to calculate the heat of solution ΔH_{soln} or the amount of energy released or absorbed in kJ mol of the substance being dissolved

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web mar 4 2014 contains water the hand warmer is struck in a manner that ruptures the inner pouch releasing the ionic salt into the water of the outer pouch the salt dissolves and the water warms this type of hand warmer tends to produce a more vigorous heat than the dry powder type of hand warmer but does not produce heat for quite as long

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web hand warmers provide a unique and fun way to study the chemistry of crystal formation and heat generation by pressing a button in a pouch which contains a supercooled solution you start a rapid exothermic heat producing crystallization in this science fair project you will determine how the starting temperature affects hand warmer chemistry

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