
Basic Transport Phenomena in Biomedical Engineering 4th Edition Fournier Solutions Manual

Chapter 1 Solutions

Problem 1.1. Use the conversion factors

$$\frac{8.314\text{J}}{\text{mol K}} \times \frac{0.23901\text{cal}}{\text{J}} = 1.987 \frac{\text{cal}}{\text{mol K}}$$

Problem 1.2. Use Equation 1.2 to convert the temperature to degrees Celsius

$$t^{\circ}\text{C} = \frac{5}{9}(t^{\circ}\text{F} - 32) = \frac{5}{9}(98.6 - 32) = 37^{\circ}\text{C}$$

Now, use Equation 1.3 to convert the temperature from degrees Celsius to Kelvin

$$T^{\circ}\text{K} = t^{\circ}\text{C} + 273.15 = 37 + 273.15 = 310.15 \text{ K}$$

Problem 1.3. Use the conversion factors

$$2 \times 10^3 \mu\text{dynes} \times \frac{\text{dynes}}{10^6 \mu\text{dynes}} \times \frac{\text{N}}{10^5 \text{dynes}} \times \frac{\text{kN}}{10^3 \text{N}} = 2 \times 10^{-10} \text{ kN}$$

Problem 1.4. Use the conversion factors

$$\frac{27 \mu\text{m}}{\text{hr}} \times \frac{10^{-6} \text{ m}}{\mu\text{m}} \times \frac{0.0006214 \text{ miles}}{\text{m}} \times \frac{\text{hr}}{60 \text{ min}} \times \frac{\text{min}}{60 \text{ sec}} = 4.66 \times 10^{-12} \frac{\text{miles}}{\text{sec}}$$

Problem 1.5. Use the conversion factors

$$48 \text{ MW} \times \frac{10^6 \text{ W}}{\text{MW}} \times \frac{\text{kg m}^2}{\text{sec}^2 \text{ W}} \times \frac{10^3 \text{ g}}{\text{kg}} \times \frac{100 \text{ cg}}{\text{g}} \times \frac{3600^3 \text{ sec}^3}{\text{hr}^3} \times \frac{100^2 \text{ cm}^2}{\text{m}^2} = 2.24 \times 10^{27} \frac{\text{cgcm}^2}{\text{hr}^3}$$

$$48 \text{ MW} \times \frac{10^6 \text{ W}}{\text{MW}} \times \frac{\text{J}}{\text{sec W}} \times \frac{\text{kJ}}{10^3 \text{ J}} \times \frac{60 \text{ sec}}{\text{min}} = 2.88 \times 10^6 \frac{\text{kJ}}{\text{min}}$$

Problem 1.6. Use the conversion factors to get both values into units of mm per second and then compare the two values for the higher value.

Transport Phenomena Biomedical Engineering Edition

R Barnett



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Table of Contents Transport Phenomena Biomedical Engineering Edition

1. Understanding the eBook Transport Phenomena Biomedical Engineering Edition
 - The Rise of Digital Reading Transport Phenomena Biomedical Engineering Edition
 - Advantages of eBooks Over Traditional Books
2. Identifying Transport Phenomena Biomedical Engineering Edition
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Transport Phenomena Biomedical Engineering Edition
 - User-Friendly Interface
4. Exploring eBook Recommendations from Transport Phenomena Biomedical Engineering Edition
 - Personalized Recommendations
 - Transport Phenomena Biomedical Engineering Edition User Reviews and Ratings
 - Transport Phenomena Biomedical Engineering Edition and Bestseller Lists
5. Accessing Transport Phenomena Biomedical Engineering Edition Free and Paid eBooks
 - Transport Phenomena Biomedical Engineering Edition Public Domain eBooks
 - Transport Phenomena Biomedical Engineering Edition eBook Subscription Services
 - Transport Phenomena Biomedical Engineering Edition Budget-Friendly Options

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 - Transport Phenomena Biomedical Engineering Edition Enhanced eBook Features
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 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Transport Phenomena Biomedical Engineering Edition
9. Balancing eBooks and Physical Books Transport Phenomena Biomedical Engineering Edition
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Transport Phenomena Biomedical Engineering Edition
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Transport Phenomena Biomedical Engineering Edition
 - Setting Reading Goals Transport Phenomena Biomedical Engineering Edition
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Transport Phenomena Biomedical Engineering Edition
 - Fact-Checking eBook Content of Transport Phenomena Biomedical Engineering Edition
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements

- Interactive and Gamified eBooks

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