

SECTION 3-1 REVIEW

CARBON COMPOUNDS

VOCABULARY REVIEW Define the following terms and provide one example for each.

1. organic compound
2. functional group
3. alcohol
4. monomer
5. polymer

MULTIPLE CHOICE Write the correct letter in the blank.

1. Organic compounds contain
a. carbon and usually other elements. c. only carbon.
b. many kinds of elements d. only carbon and hydrogen.
except carbon.
2. The number of covalent bonds a carbon atom can form with other atoms is
a. 1. b. 2. c. 4. d. 8.
3. A covalent bond formed when two atoms share two pairs of electrons is called a
a. single bond. b. double bond. c. triple bond.
d. quadruple b. bond.
4. The breakdown of a polymer involves
a. hydrolysis. c. the breaking of hydrogen bonds.
b. a condensation reaction. d. the breaking of ionic bonds.
5. ATP releases energy when
a. it undergoes a condensation reaction. c. a phosphate group is added to it.
b. a hydroxyl group is added to it. d. a phosphate group is removed from it.

Carbon Compounds Section 3 1 Review Answers

**Ralph M. Feather, Susan Leach
Snyder, Dale T. Hesser**



Carbon Compounds Section 3 1 Review Answers:

Chemistry 'O' Level Rex M. Heyworth, 2007 Holt Chemistry Ralph Thomas Myers, 2004 *McGraw-Hill's SAT Subject Test Biology E/M, 3rd Edition* Stephanie Zinn, 2012-02-03 Expert guidance on the Biology E M exam Many colleges and universities require you to take one or more SAT II Subject Tests to demonstrate your mastery of specific high school subjects McGraw Hill's SAT Subject Test Biology E M is written by experts in the field and gives you the guidance you need to perform at your best This book includes 4 full length sample tests updated for the latest test formats two practice Biology E exams and two practice Biology M exams 30 top tips to remember for test day Glossary of tested biology terms How to decide whether to take Biology E or Biology M Diagnostic test to pinpoint strengths and weaknesses Sample exams exercises and problems designed to match the real tests in content and level of difficulty Step by step review of all topics covered on the two exams In depth coverage of the laboratory experiment questions that are a major part of the test **Chemistry:**

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Modern Chemistry Holt Rinehart & Winston, Holt, Rinehart and Winston Staff, 2001 Chemistry Richard Post, Chad

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Merrill Earth Science Ralph M. Feather, Susan Leach Snyder, Dale T. Hesser, 1995 *Science in Your World: Teacher edition* Jay K. Hackett, 1991 **Science Spectrum** Holt Rinehart & Winston, Holt, Rinehart and Winston Staff, 2003-03 Organic Chemistry L. G. Wade, 2010 Organized around functional groups this book incorporates problem solving help orientation features and complete discussions of mechanisms Acid Base Chemistry Lewis Structures Bronsted Electron Structure shell orbitals magnetic shielding Bonding formation patterns polarity MO Resonance Stereochemistry MO Theory Conformational analysis Thermodynamics Kinetics Reaction Coordinate diagrams Chirality Regioselectivity Synthesis Aromaticity Carbonyl chemistry A comprehensive reference for chemistry professionals **Glencoe Science** , 2002 *Glencoe Science* McGraw-Hill Staff, 2001-08 **General Applied Chemistry** Stanley E. Manahan, 1978 **The Electrical Review** , 1924 Essentials of Biology Holt Rinehart & Winston, 1998 **Physical Science Junior High School Science Series 1986** , 1986-06 **Addison-Wesley Science Insights** , 1996 *Science Insights* , 1999 **Foundations of Anatomy and Physiology** Frederic Martini, George Karleskint, 1998 THE STORY It is Christmas in New York but for two young lovers Jim and Della the prospects are bleak as both are out of work and penniless But as those familiar with the famous O Henry story are aware their dilemma is solved when both part with th

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