

# CHEMISTRY



**Atomic Structure**

Atoms are the smallest particles that cannot be created or destroyed in a chemical reaction. They are made up of three sub-particles: electrons, protons, and neutrons.

**Atomic Number**

The atomic number of an element is the number of protons present in the nucleus of an atom of that element. It is denoted by the symbol  $Z$ .

**Atomic Mass**

The atomic mass of an element is the sum of the number of protons and neutrons present in the nucleus of an atom of that element. It is denoted by the symbol  $A$ .

**Atomic Structure**

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Element	Atomic Number	Atomic Mass
Hydrogen	1	1
Helium	2	4
Lithium	3	7
Boron	5	11
Carbon	6	12
Nitrogen	7	14
Oxygen	8	16
Fluorine	9	19
Neon	10	20

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# Quick Study Laminated Reference Guides

**Hussin A.Rothana**



**Quick Study Laminated Reference Guides:**

## **Quick Study Laminated Reference Guides** Book Review: Unveiling the Power of Words

In some sort of driven by information and connectivity, the power of words has are more evident than ever. They have the ability to inspire, provoke, and ignite change. Such is the essence of the book **Quick Study Laminated Reference Guides**, a literary masterpiece that delves deep in to the significance of words and their impact on our lives. Published by a renowned author, this captivating work takes readers on a transformative journey, unraveling the secrets and potential behind every word. In this review, we shall explore the book is key themes, examine its writing style, and analyze its overall effect on readers.

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