

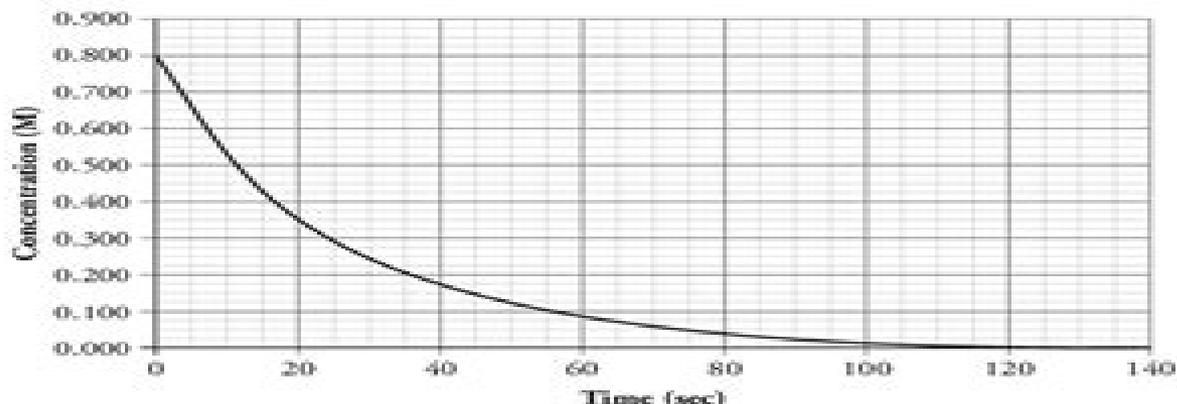
Rate of Reaction

How is the speed of a reaction measured?

Why?

Chemical reactions occur at different speeds. Some are almost instantaneous. Others require patience. For example, rust can form on iron in just a few days or over a period of months depending on the conditions. In order to study the factors that change the speed of a reaction, we must first develop an understanding of how the rate of reaction is monitored during a reaction.

Model 1 – Concentration versus Time Graph



1. The graph in Model 1 illustrates how the concentration of a species in a chemical reaction changes over time.
 - a. What unit is the concentration measured in?
 - b. What unit is the time measured in?
2. Consider the data in Model 1.
 - a. What was the concentration of the species when the chemical reaction was initiated?
 - b. Did the concentration of the species increase or decrease over time?
 - c. Was the species a reactant or product in the reaction? Justify your reasoning.

Rate Of Reaction 1 Answers Key

**William R. Robinson, Jerome D.
Odom, Henry Fuller Holtzclaw**

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