

## **ZOOL Chapter 2: The Chemical Level of Organization**

### ***2.1 How Matter is Organized***

- Chemistry – science of structure and interactions of matter
  - o Matter – anything that occupies space and mass
  - o Mass – the amount of matter in an object
  - o Weight – force of gravity acting on matter

#### **Chemical Elements**

- Three types of matter: solids, liquids, and gasses
- Chemical elements
  - o Make up all forms of living matter
  - o Chemical symbol: N, O, Au, Fe
- Major elements: O, C, H, N
  - o Makes up 96% of body's mass
- Lesser elements (8)
- Trace elements (14)
  - o Exist in trace amounts

#### **Structure of Atoms**

- Atoms – smallest unit of matter
- Subatomic particles:
  - o Electrons
  - o Neutrons
  - o Protons

#### **Atomic Mass and Number**

- Atomic number = # of protons
- Mass number = # of protons + neutrons
- Isotopes are atoms of an element that have different amounts of neutrons
  - o Radioactive isotopes are unstable
    - As they decay, they emit radiation

#### **Atomic Mass**

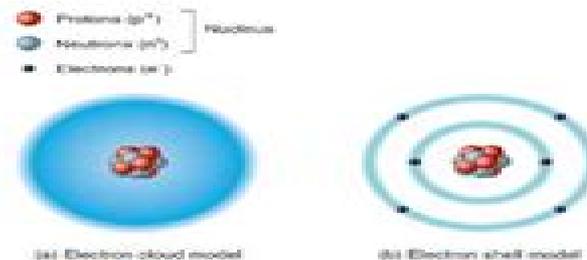
- Atomic mass unit
- A neutron has 1.008 amu
- Protons have 1.007 amu

#### **Ions, Molecules, and Compounds**

- Ion is an atom with a positive or negative charge
- Ionization – process of giving up or gaining electrons
- Molecules – two atoms SHARE electrons
- Compounds – substance that contains two or more atoms of different elements
- Free radical – atom or group of atoms with an unpaired electron

### ***2.2 Chemical Bonds***

- Chemical bonds – the forces that holds atoms of a molecule/compound together
- Valence shell – outermost energy level
- Octet rule – electrons seek to fill out all energy level



# Anatomy And Physiology Chapter 2

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Findlay, Cheng Hean Lo**

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