

CHEMISTRY PAPER 1 (THEORY)

Maximum Marks: 70

Time Allowed: Three hours

*Candidates are allowed additional 15 minutes for only reading the paper.
They must NOT start writing during this time.*

This paper is divided into four sections – A, B, C and D.

Answer all questions.

Section A consists of one question having sub-parts of one mark each.

Section B consists of ten questions of two marks each.

Section C consists of seven questions of three marks each, and

Section D consists of three questions of five marks each.

*Internal choices have been provided in one question each in Section B,
Section C and Section D.*

*All working, including rough work, should be done on the same sheet as, and adjacent to the rest
of the answer.*

The intended marks for questions or parts of questions are given in brackets [].

Balanced equations must be given wherever possible and diagrams where they are helpful.

When solving numerical problems, all essential working must be shown.

In working out problems, use the following data:

*Gas constant $R = 1.987 \text{ cal deg}^{-1} \text{ mol}^{-1} = 8.314 \text{ JK}^{-1} \text{ mol}^{-1}$
 $= 0.0821 \text{ dm}^3 \text{ atm K}^{-1} \text{ mol}^{-1}$*

$1 \text{ l atm} = 1 \text{ dm}^3 \text{ atm} = 101.3 \text{ J}$, $1 \text{ Faraday} = 96500 \text{ coulombs}$.

Avogadro's number = 6.023×10^{23} .

SECTION A – 14 MARKS

Question 1

(A) Fill in the blanks by choosing the appropriate word(s) from those given in the [4×1]
brackets:

[stable, low, aldehyde, unstable, 6, 4, ethane, Clemmensen's, 2, 3, carboxylic acid,
high, propane, Rosenmund's]

- (i) The primary alcohols are easily oxidised first into _____ and then into _____.
- (ii) The intermediate activated complex in a chemical reaction is highly _____ due to _____ energy.
- (iii) The coordination number and oxidation state of the complex $\text{K}_4[\text{Fe}(\text{CN})_6]$ are _____ and _____ respectively.
- (iv) Propanone on reaction with zinc-amalgam in presence of conc. HCl gives _____ and the reaction is known as _____ reduction.

This Paper consists of 8 printed pages.

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KJ Lindholm-Leary



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1. Understanding the eBook Bsc Sem 3 Question Paper Chemistry Nolcom
 - The Rise of Digital Reading Bsc Sem 3 Question Paper Chemistry Nolcom
 - Advantages of eBooks Over Traditional Books
2. Identifying Bsc Sem 3 Question Paper Chemistry Nolcom
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Bsc Sem 3 Question Paper Chemistry Nolcom
 - User-Friendly Interface
4. Exploring eBook Recommendations from Bsc Sem 3 Question Paper Chemistry Nolcom
 - Personalized Recommendations
 - Bsc Sem 3 Question Paper Chemistry Nolcom User Reviews and Ratings
 - Bsc Sem 3 Question Paper Chemistry Nolcom and Bestseller Lists
5. Accessing Bsc Sem 3 Question Paper Chemistry Nolcom Free and Paid eBooks
 - Bsc Sem 3 Question Paper Chemistry Nolcom Public Domain eBooks
 - Bsc Sem 3 Question Paper Chemistry Nolcom eBook Subscription Services
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 - Bsc Sem 3 Question Paper Chemistry Nolcom Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Bsc Sem 3 Question Paper Chemistry Nolcom
 - Highlighting and Note-Taking Bsc Sem 3 Question Paper Chemistry Nolcom
 - Interactive Elements Bsc Sem 3 Question Paper Chemistry Nolcom
8. Staying Engaged with Bsc Sem 3 Question Paper Chemistry Nolcom
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Bsc Sem 3 Question Paper Chemistry Nolcom
9. Balancing eBooks and Physical Books Bsc Sem 3 Question Paper Chemistry Nolcom
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Bsc Sem 3 Question Paper Chemistry Nolcom
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Bsc Sem 3 Question Paper Chemistry Nolcom
 - Setting Reading Goals Bsc Sem 3 Question Paper Chemistry Nolcom
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Bsc Sem 3 Question Paper Chemistry Nolcom
 - Fact-Checking eBook Content of Bsc Sem 3 Question Paper Chemistry Nolcom
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements

- Interactive and Gamified eBooks

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