

1984 Advanced Placement Exam

Part I Multiple Choice

Note: For all questions referring to solutions, assume that the solvent is water unless otherwise stated.

Directions: Each set of lettered choices below refers to the numbered statements immediately following it. Select the one lettered choice that best fits each statement and then blacken the corresponding space on the answer sheet. A choice may be used once, more than once, or not at all in each set.

Questions 1-3

- (A) F
- (B) S
- (C) Mg
- (D) Ar
- (E) Mn

1. Forms monatomic ions with 2- charge in solutions
2. Forms a compound having the formula KXO_4
3. Forms oxides that are common air pollutants and that yield acidic solution in water

Questions 4-7

- (A) Hydrofluoric acid
- (B) Carbon dioxide
- (C) Aluminum hydroxide
- (D) Ammonia
- (E) Hydrogen peroxide

4. Is a good oxidizing agent
5. Is used to etch glass chemically
6. Is used extensively for the production of fertilizers
7. Has amphoteric properties

Questions 8-9

- (A) A network solid with covalent bonding
- (B) A molecular solid with zero dipole moment
- (C) A molecular solid with hydrogen bonding
- (D) An ionic solid
- (E) A metallic solid

8. Solid ethyl alcohol, C_2H_5OH
9. Silicon dioxide, SiO_2

Questions 10-13

- (A) CO_3^{2-}
- (B) $Cr_2O_7^{2-}$
- (C) NH_4^+
- (D) Ba^{2+}
- (E) Al^{3+}

Assume that you have an "unknown" consisting of an aqueous solution of a salt that contains one of the ions listed above. Which ion must be absent on the basis of each of the following observations of the "unknown"?

10. The solution is colorless
11. The solution gives no apparent reaction with dilute hydrochloric acid.
12. No odor can be detected when a sample of the solution is added drop by drop to a warm solution of sodium hydroxide.
13. No precipitate is formed when a dilute solution of H_2SO_4 is added to a sample of the solution.

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John W. Crum



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Advanced Placement Examination in Chemistry Frederick J. Rowe, 1990 New for 1990 this is the only preparation guide available for this exam Includes analyses of past tests review of chemistry topics and samples of all multiple choice and free response problems

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The Cumulative Book Index, 1985 A world list of books in the English language

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CQ, 1992

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