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web choose the best answer from the options that follow each question 32 air is about 78 nitrogen 21 oxygen and 1 other gases after all the oxygen is removed from a sample of air in a glass tube and the temperature remains constant the pressure exerted by the remaining air a does not change b

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web temperature and pressure hints density g l use the ideal gas law to solve for moles first a 0 35 g l b 2 14 g l c 2 86 g l first use the ideal gas law and recall that density is g l 101 3 x 1 l n x 8 31 x 273 therefore n 0 045 moles density is g l so convert moles to grams by multiplying by the molar mass of so 2

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