



# Apple Music Update

**Yijin Wang**



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web the arrhenius theory of acids and bases acids are substances that produce hydrogen ions in solution bases are

substances that produce hydroxide ions in solution neutralization happens because hydrogen ions and hydroxide ions react to produce water  $\text{H}^+ + \text{OH}^- \rightarrow \text{H}_2\text{O}$  hydrochloric

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web the  $\text{A}^-$  ion is a base because it is accepting a proton from the  $\text{H}_3\text{O}^+$  the reversible reaction contains two acids and two bases we think of them in pairs called conjugate pairs when the acid  $\text{HA}$  loses a proton it forms a base  $\text{A}^-$  when the base  $\text{A}^-$  accepts a proton back again it obviously reforms the acid  $\text{HA}$

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