

**TEACHING TRANSPARENCY WORKSHEET**

# Types of Matter

**Use with Chapter 3,  
Section 3.4**

1. Into what two broad classes can all matter be divided?

\_\_\_\_\_

2. What is the difference between a mixture and a pure substance?

\_\_\_\_\_

3. What is the difference between a compound and an element?

\_\_\_\_\_

4. Can a compound be a heterogeneous mixture? Explain your answer by referring to the diagram.

\_\_\_\_\_

\_\_\_\_\_

5. A list of compounds and elements is given below. Circle the substances that are elements.

- |       |          |          |
|-------|----------|----------|
| gold  | aluminum | water    |
| sugar | oxygen   | platinum |
| salt  | chlorine | brass    |

6. How can you tell the difference between a homogeneous mixture and a heterogeneous mixture?

\_\_\_\_\_

\_\_\_\_\_

7. Label each mixture below as either homogeneous or heterogeneous.

- |  |                       |
|--|-----------------------|
| a. air _____                           | e. finger paint _____ |
| b. clay _____                          | f. vinegar _____      |
| c. homemade lemonade (with pulp) _____ | g. soil _____         |
| d. oatmeal raisin cookie _____         |                       |

8. List three methods that are commonly used to separate mixtures into their component substances.

\_\_\_\_\_

\_\_\_\_\_

# Teaching Transparency Answers Chapter 18

**John S. Phillips, Cheryl Wistrom**



## Teaching Transparency Answers Chapter 18:

*American Republic to 1877 Teaching Transparency Sampler 2003* McGraw-Hill Staff,2002-09      **American Republic Since 1877 Teaching Transparency Sampler 2003** McGraw-Hill Staff,2002-08      *Instructor's Manual and Transparency Masters* David J. Rachman,1985      **Ch 1-12 W/Wrkppr-Acct** Warren/Reeve/Fess,Carl S. (Carl S. Warren) Warren,James M. Reeve,Philip E. Fess,2005-06      **Instructor's Guide for Nutrition and Diet Therapy** Carroll A. Lutz,2001

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