

## Standard 10 year

- Most popular
- 10 year term
- Equal monthly payments
- Good for those with student loans less than annual salary

## Income-Based Repayment (IBR)

- Capped at 10% of your income
- 20 to 25 year term
- Payment must be less than payment under the Standard Repayment Plan.
- Forgiven loans subject to taxes

## Saving on a Valuable Education (SAVE)

- 5% of your income
- Undergrad: 20 Year Term
- Grad: 25 Year Term
- Loan forgiveness as early as 10 years
- Good for most borrowers

 THE COLLEGE INVESTOR

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# Student Loan Repayment Plans

## Extended

- Extra time to pay
- 10 year to 30 year term
- Good for those with student loan balance between 1-1.5 times annual earnings

## Pay As You Earn (PAYE)

- 10% of your income
- 20 year term
- Loan forgiven after term
- Capped if 10% of income more than the payment under the Standard Repayment Plan.
- Payment must be less than payment under the Standard Repayment Plan.
- Forgiven loans subject to taxes

## Graduated

- Income-driven
- 10 year term
- 30 year term for direct consolidation loans
- Payments start low
- Adjustable as income rises
- Minimum payment increases every two years

## Income-Contingent Repayment (ICR)

- Pay the lesser of either 20% of income or total payment over 12 years
- 12 year-25 year term
- Forgiven amount subject to taxes
- Direct student loans qualify



PSLF Eligible

# Student Loan Repayment Best

**Joacim Rocklöv**



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web 11 use avogadro s law to determine the number of moles of butane gas assume that butane is an ideal gas and that one mole has a volume of 22 4 l at stp v 1 n 1 v 2 n 2 0 0896 l n 22 4 l 1 mole n 0 0040 moles 12 experimental molar mass g mole of butane 0 23 g 0 0040 moles 58 g mole 13 the molecular formula of butane is c 4 h 10

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web nov 3 2020 in this experiment the moles and mass will be required to determine the molar mass of butane in a butane lighter experimental to begin the experiment the mass of a butane lighter was measured before using the butane gas to deplete the water to the 80ml mark on a graduated cylinder

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web the pressure of the gas the pressure of the gas was calculated using dalton s law of partial pressures by subtracting the pressure of the water vapor from the total pressure which in this case was the atmospheric pressure 760mm hg the pressure of the water vapor was found by using the temperature of the water

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web instructions choose an answer and hit next you will receive your score and answers at the end question 1 of 3 butane and isobutane are constitutional isomers which of the following best

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web use the combined gas law to determine the volume in l of butane at stp use avogadro s law to determine the number of moles of butane gas assume that butane is an ideal gas and that one mole has a volume of 22 l at stp experimental molar mass g mole of butane the molecular formula of butane is c 4 h 10

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web although the most convenient gas for this is butane other gases may be available in similar small cylinders the collection of a gas sample and the weighing of the gas cylinder before and after this should take about 5 10 minutes as a demonstration