



Carbon and its compounds

- Carbon is an element
- It is a non-metal
- The amount of carbon present in the earth's crust and atmosphere is very small
- All the living things, plants and animals are made up of carbon based compounds which are called organic compounds

Carbon always forms covalent bonds

- It is not possible to remove 4 electrons from a carbon atom. It is also not possible to add as many as 4 electrons to a carbon atom.
- * Since carbon atoms can achieve the inert gas electron arrangement only by the sharing of electrons, therefore, carbon always forms covalent bonds.

Carbon is Tetravalent

One carbon atom requires 4 electrons to achieve the eight-electron inert gas structure, therefore the valency of carbon is 4.

Self Combination.

The property of self combination of carbon atoms to form long chains is useful to us because it gives rise to an extremely large number of carbon compounds



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