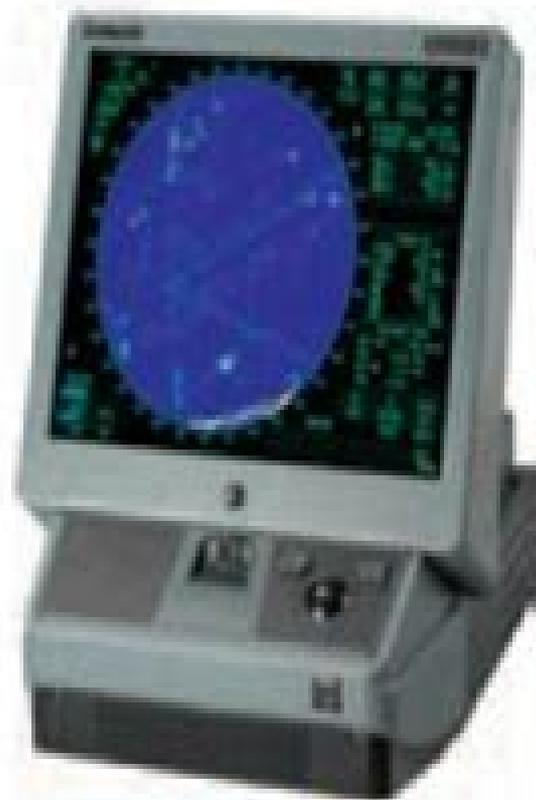




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web write the oxidation and reduction reactions for each redox reaction the first one is done for you 23 $\text{Fe} + \text{Co} \rightarrow \text{Fe} + \text{Co} + 2\text{e}^-$ oxidation $\text{Co} + 2\text{e}^- \rightarrow \text{Co} + 2\text{e}^-$ reduction $\text{Fe} + 2\text{e}^- \rightarrow \text{Fe} + 2\text{e}^-$ 24 3 $\text{Ag} + \text{Ni} \rightarrow \text{Ag} + \text{Ni} + 3\text{e}^-$ oxidation $\text{Ni} + 3\text{e}^- \rightarrow \text{Ni} + 3\text{e}^-$ reduction 25 $\text{Cu} + \text{Pb} \rightarrow \text{Cu} + \text{Pb} + 2\text{e}^-$ oxidation $\text{Pb} + 2\text{e}^- \rightarrow \text{Pb} + 2\text{e}^-$ reduction 26 O_2

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