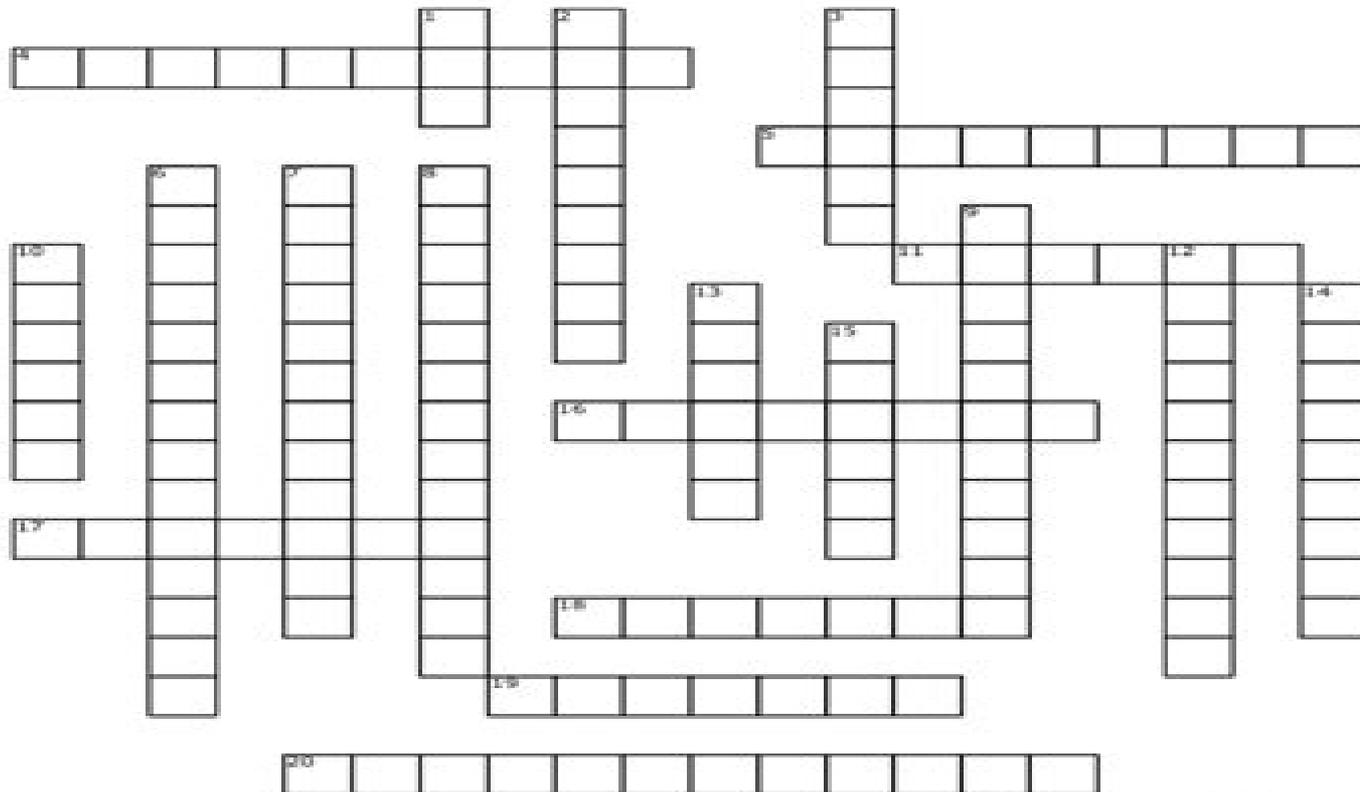


# Gas Laws Crossword - Abigahile Bruny



## Across

- 4.** Water in the form of a gas  
**5.** In Boyle's Law, at a constant temperature, the volume of the gas increases as the pressure \_\_\_\_\_  
**11.** \_\_\_\_\_ is the unit measurement for temperature  
**16.** A force pressing or expanding on another.  
**17.** \_\_\_\_\_ law states the direct relationship between the volume and temperature of a gas at constant pressure.  
**18.** A relationship in which one variable decreases when \_\_\_\_\_  
**19.** A real gas is a gas that does not behave as an ideal gas because of the interactions between gas molecules.  
**20.** Temperature of -273.15 C given a value of zero in the Kelvin Scale

## Down

- 1.** A substance or matter in a state in which it will expand freely to fill the whole of a container, having no fixed shape.  
**2.** The Kinetic \_\_\_\_\_ Theory explains the behavior of matter in terms of particles that are in constant motion  
**3.** The relationship between two variables that shows the same effect.  
**6.** The Law that describes the relationship among the pressure, temperature, and volume of a gas.  
**7.** The volume and amount (moles) of gas are directly proportional  $\Rightarrow V_1n_2 = V_2n_1$   
**8.** When the volume is constant, as pressure increases, temperature increases

- 9.** The average kinetic energy of individual particles.

- 10.** Heating up the molecules of a gas allow the particles to move \_\_\_\_\_  
**12.** The relationship  $PV = nRT$ , which describes the behavior of an ideal gas.  
**13.** The unit of measurement used to describe volume in Ideal Gas Law  
**14.** An instrument that measures atmospheric pressure.  
**15.** \_\_\_\_\_ is the SI unit of pressure

# Gas Laws Crossword Answer Key

**RM Cervero**



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