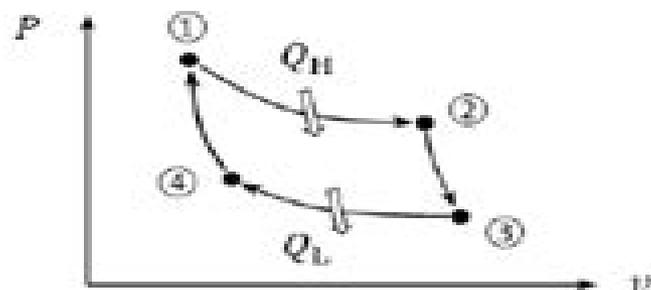


Written Problem

The Carnot cycle is an idealization that involves four processes as shown (see also Figure 6–37 in textbook). As each of the processes is reversible, the entire cycle is also reversible.



- ① → ② : Isothermal expansion at $T = T_H$
- ② → ③ : Adiabatic expansion (no heat transfer)
- ③ → ④ : Isothermal compression at $T = T_L$
- ④ → ① : Adiabatic compression (no heat transfer)

- (a) Suppose the Carnot cycle is carried out in a closed system with an ideal gas as working fluid. Show that

$$Q_H = mRT_H \ln(V_2/V_1)$$

where m is the mass of the ideal gas. (Recall that the internal energy of an ideal gas depends *only* on the temperature T .)

You are now provided with more specifics: The ideal gas is 2 g of air. The Carnot cycle operates between $T_H = 500$ K and $T_L = 300$ K, with $V_2 = 2V_1$.

- (b) What is the amount of heat Q_L (kJ) that is transferred to the cold reservoir in every cycle?
- (c) Determine W_{net} (kJ), the net amount of work that is produced per cycle.

Note that W_{net} can be obtained by summing the $\int P \, dV$ contribution from each of the four processes. This approach, however, is not recommended as there are much easier ways to get to the answer.

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TD Snyder



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Basics Of Statistical Physics (Third Edition) Harald J W Muller-kirsten, 2022-03-16 Statistics links microscopic and macroscopic phenomena and requires for this reason a large number of microscopic elements like atoms The results are values of maximum probability or of averaging This introduction to statistical physics concentrates on the basic principles and attempts to explain these in simple terms supplemented by numerous examples These basic principles include the difference between classical and quantum statistics a priori probabilities as related to degeneracies the vital aspect of indistinguishability as compared with distinguishability in classical physics the differences between conserved and non conserved elements the different ways of counting arrangements in the three statistics Maxwell Boltzmann Fermi Dirac Bose Einstein the difference between maximization of the number of arrangements of elements and averaging in the Darwin Fowler method Significant applications to solids radiation and electrons in metals are treated in separate chapters as well as Bose Einstein condensation In this latest edition apart from a general revision the topic of thermal radiation has been expanded with a new section on black bodies and an additional chapter on black holes Other additions are more examples with applications of statistical mechanics in solid state physics and superconductivity Throughout the presentation the introduction carries almost all details for calculations

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