
IGCSE CHEMISTRY NOTES

Syllabus

[IGCSE Chemistry \(0620\) Syllabus](#)

The particulate nature of matter

- Distinguishing properties

| SOLID | LIQUID | GAS |
|--|--|--|
| <ul style="list-style-type: none">• Strong forces of attraction between particles• Fixed pattern (lattice)• Atoms vibrate but can't change position \therefore fixed volume and shape | <ul style="list-style-type: none">• Weaker attractive forces than solids• No fixed pattern, liquids take up the shape of their container• Particles slide past each other. | <ul style="list-style-type: none">• Almost no intermolecular forces• Particles far apart, and move quickly• Collide with each other and bounce in all directions |

- Structure
 - Solids - Particles are arranged in a fixed particle or lattice. They cannot leave their positions. They have a to and fro motion called vibration
 - Liquid - The particles in a liquid can move around and slide past each other. They are close but have no lattice. 3R motion - Random, Rapid, Rectilinear
 - Gases - There are almost no forces holding them together. 3R motion - Random, Rapid, Rectilinear

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Hussin A.Rothana



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Table of Contents Cambridge Chemistry Notes

1. Understanding the eBook Cambridge Chemistry Notes
 - The Rise of Digital Reading Cambridge Chemistry Notes
 - Advantages of eBooks Over Traditional Books
2. Identifying Cambridge Chemistry Notes
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Cambridge Chemistry Notes
 - User-Friendly Interface
4. Exploring eBook Recommendations from Cambridge Chemistry Notes
 - Personalized Recommendations
 - Cambridge Chemistry Notes User Reviews and Ratings
 - Cambridge Chemistry Notes and Bestseller Lists
5. Accessing Cambridge Chemistry Notes Free and Paid eBooks
 - Cambridge Chemistry Notes Public Domain eBooks
 - Cambridge Chemistry Notes eBook Subscription Services
 - Cambridge Chemistry Notes Budget-Friendly Options
6. Navigating Cambridge Chemistry Notes eBook Formats

- ePub, PDF, MOBI, and More
 - Cambridge Chemistry Notes Compatibility with Devices
 - Cambridge Chemistry Notes Enhanced eBook Features
7. Enhancing Your Reading Experience
- Adjustable Fonts and Text Sizes of Cambridge Chemistry Notes
 - Highlighting and Note-Taking Cambridge Chemistry Notes
 - Interactive Elements Cambridge Chemistry Notes
8. Staying Engaged with Cambridge Chemistry Notes
- Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Cambridge Chemistry Notes
9. Balancing eBooks and Physical Books Cambridge Chemistry Notes
- Benefits of a Digital Library
 - Creating a Diverse Reading Collection Cambridge Chemistry Notes
10. Overcoming Reading Challenges
- Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Cambridge Chemistry Notes
- Setting Reading Goals Cambridge Chemistry Notes
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Cambridge Chemistry Notes
- Fact-Checking eBook Content of Cambridge Chemistry Notes
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
- Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
- Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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