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# Basic Transport Phenomena in Biomedical Engineering 4th Edition Fournier Solutions Manual

## Chapter 1 Solutions

**Problem 1.1.** Use the conversion factors

$$\frac{8.314\text{J}}{\text{mol K}} \times \frac{0.23901\text{cal}}{\text{J}} = 1.987 \frac{\text{cal}}{\text{mol K}}$$

**Problem 1.2.** Use Equation 1.2 to convert the temperature to degrees Celsius

$$t^{\circ}\text{C} = \frac{5}{9}(t^{\circ}\text{F} - 32) = \frac{5}{9}(98.6 - 32) = 37^{\circ}\text{C}$$

Now, use Equation 1.3 to convert the temperature from degrees Celsius to Kelvin

$$T^{\circ}\text{K} = t^{\circ}\text{C} + 273.15 = 37 + 273.15 = 310.15 \text{ K}$$

**Problem 1.3.** Use the conversion factors

$$2 \times 10^3 \mu\text{dynes} \times \frac{\text{dynes}}{10^6 \mu\text{dynes}} \times \frac{\text{N}}{10^5 \text{dynes}} \times \frac{\text{kN}}{10^3 \text{N}} = 2 \times 10^{-10} \text{ kN}$$

**Problem 1.4.** Use the conversion factors

$$\frac{27 \mu\text{m}}{\text{hr}} \times \frac{10^{-6} \text{ m}}{\mu\text{m}} \times \frac{0.0006214 \text{ miles}}{\text{m}} \times \frac{\text{hr}}{60 \text{ min}} \times \frac{\text{min}}{60 \text{ sec}} = 4.66 \times 10^{-12} \frac{\text{miles}}{\text{sec}}$$

**Problem 1.5.** Use the conversion factors

$$48 \text{ MW} \times \frac{10^6 \text{ W}}{\text{MW}} \times \frac{\text{kg m}^2}{\text{sec}^2 \text{ W}} \times \frac{10^3 \text{ g}}{\text{kg}} \times \frac{100 \text{ cg}}{\text{g}} \times \frac{3600^3 \text{ sec}^3}{\text{hr}^3} \times \frac{100^2 \text{ cm}^2}{\text{m}^2} = 2.24 \times 10^{27} \frac{\text{cgcm}^2}{\text{hr}^3}$$

$$48 \text{ MW} \times \frac{10^6 \text{ W}}{\text{MW}} \times \frac{\text{J}}{\text{sec W}} \times \frac{\text{kJ}}{10^3 \text{ J}} \times \frac{60 \text{ sec}}{\text{min}} = 2.88 \times 10^6 \frac{\text{kJ}}{\text{min}}$$

**Problem 1.6.** Use the conversion factors to get both values into units of mm per second and then compare the two values for the higher value.

# Basic Transport Phenomena In Biomedical Engineering

**Marcel A. Müller**



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## **Table of Contents Basic Transport Phenomena In Biomedical Engineering**

1. Understanding the eBook Basic Transport Phenomena In Biomedical Engineering
  - The Rise of Digital Reading Basic Transport Phenomena In Biomedical Engineering
  - Advantages of eBooks Over Traditional Books
2. Identifying Basic Transport Phenomena In Biomedical Engineering
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Basic Transport Phenomena In Biomedical Engineering
  - User-Friendly Interface
4. Exploring eBook Recommendations from Basic Transport Phenomena In Biomedical Engineering
  - Personalized Recommendations
  - Basic Transport Phenomena In Biomedical Engineering User Reviews and Ratings
  - Basic Transport Phenomena In Biomedical Engineering and Bestseller Lists
5. Accessing Basic Transport Phenomena In Biomedical Engineering Free and Paid eBooks
  - Basic Transport Phenomena In Biomedical Engineering Public Domain eBooks
  - Basic Transport Phenomena In Biomedical Engineering eBook Subscription Services
  - Basic Transport Phenomena In Biomedical Engineering Budget-Friendly Options

6. Navigating Basic Transport Phenomena In Biomedical Engineering eBook Formats
  - ePub, PDF, MOBI, and More
  - Basic Transport Phenomena In Biomedical Engineering Compatibility with Devices
  - Basic Transport Phenomena In Biomedical Engineering Enhanced eBook Features
7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Basic Transport Phenomena In Biomedical Engineering
  - Highlighting and Note-Taking Basic Transport Phenomena In Biomedical Engineering
  - Interactive Elements Basic Transport Phenomena In Biomedical Engineering
8. Staying Engaged with Basic Transport Phenomena In Biomedical Engineering
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Basic Transport Phenomena In Biomedical Engineering
9. Balancing eBooks and Physical Books Basic Transport Phenomena In Biomedical Engineering
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Basic Transport Phenomena In Biomedical Engineering
10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
11. Cultivating a Reading Routine Basic Transport Phenomena In Biomedical Engineering
  - Setting Reading Goals Basic Transport Phenomena In Biomedical Engineering
  - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Basic Transport Phenomena In Biomedical Engineering
  - Fact-Checking eBook Content of Basic Transport Phenomena In Biomedical Engineering
  - Distinguishing Credible Sources
13. Promoting Lifelong Learning
  - Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
14. Embracing eBook Trends
  - Integration of Multimedia Elements

- Interactive and Gamified eBooks

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