

3 • Reactions in Aqueous Solution**STUDY QUESTIONS AND PROBLEMS**

1. Classify each of the following solutes as a strong electrolyte, weak electrolyte, or nonelectrolyte:

sugar	sodium hydroxide
common salt (NaCl)	hydrochloric acid
alcohol	copper sulfate
acetic acid	carbonic acid

2. Predict the solubility of the following salts:

- sodium sulfate
- potassium chromate
- silver bromide
- nickel (II) hydroxide
- aluminum nitrate
- barium sulfide
- ammonium acetate
- strontium iodide

3. Write the ions that are produced when the following substances dissolve in water:

- $\text{Mg}(\text{OH})_2$
- K_2SO_4
- NaHCO_3
- $(\text{NH}_4)_3\text{PO}_4$
- NaClO

4. Predict whether or not the following reactions will lead to a precipitate. Write complete and net ionic equations for all the reactions.

- a. potassium chromate and lead acetate
- b. silver perchlorate and ammonium chloride
- c. potassium carbonate and copper acetate
- d. sodium fluoride and magnesium iodide
- e. barium nitrate and potassium sulfate

Reactions In Aqueous Solution Worksheet Answers

Michael Hoelscher



Reactions In Aqueous Solution Worksheet Answers:

Reactions In Aqueous Solution Worksheet Answers Book Review: Unveiling the Power of Words

In a world driven by information and connectivity, the power of words has become more evident than ever. They have the ability to inspire, provoke, and ignite change. Such could be the essence of the book **Reactions In Aqueous Solution Worksheet Answers**, a literary masterpiece that delves deep to the significance of words and their affect our lives. Compiled by a renowned author, this captivating work takes readers on a transformative journey, unraveling the secrets and potential behind every word. In this review, we will explore the book is key themes, examine its writing style, and analyze its overall effect on readers.

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