

## Qualitative Analysis and Chemical Bonding

AP Chemistry Big Idea 1: Identification of  
An Inferred Property (1)

### Introduction

Learning the qualitative properties of substances and the chemical bonds that hold them together are essential to understanding their behavior. This activity provides students with the opportunity to explore the relationship between chemical bonding and the properties of substances.

### Concepts

- 1. Chemical bonding
- 2. Chemical bonding
- 3. Chemical bonding
- 4. Chemical bonding
- 5. Chemical bonding

### Background

Chemical bonding is the force that holds atoms together. The type of chemical bond is related to the electronegativity of the atoms. The electronegativity of an atom is a measure of its ability to attract electrons. The electronegativity of an atom is related to its position in the periodic table. The electronegativity of an atom increases from left to right across the periodic table and from bottom to top.

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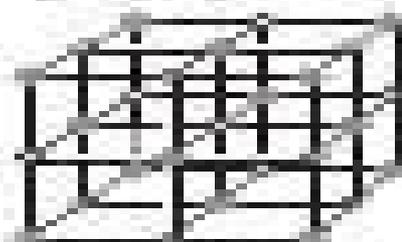


Figure 1: Crystal Lattice Structure

- 1. Crystal lattice
- 2. Crystal lattice

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