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**L. R. Sharma, B. R. Puri, Madan S.  
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Indian Universities L. R. Sharma, B. R. Puri, Madan S. Pathania, 1986      *Text Book of Physical Chemistry* Kewal K. Sharma, L. K. Sharma, 1977      **A Textbook of Physical Chemistry - Volume 1** Mandeep Dalal, 2018-01-01 An advanced level textbook of physical chemistry for the graduate B Sc and postgraduate M Sc students of Indian and foreign universities This book is a part of four volume series entitled A Textbook of Physical Chemistry Volume I II III IV CONTENTS Chapter 1 Quantum Mechanics I Postulates of quantum mechanics Derivation of Schrodinger wave equation Max Born interpretation of wave functions The Heisenberg s uncertainty principle Quantum mechanical operators and their commutation relations Hermitian operators elementary ideas quantum mechanical operator for linear momentum angular momentum and energy as Hermitian operator The average value of the square of Hermitian operators Commuting operators and uncertainty principle  $x E$  Schrodinger wave equation for a particle in one dimensional box Evaluation of average position average momentum and determination of uncertainty in position and momentum and hence Heisenberg s uncertainty principle Pictorial representation of the wave equation of a particle in one dimensional box and its influence on the kinetic energy of the particle in each successive quantum level Lowest energy of the particle Chapter 2 Thermodynamics I Brief resume of first and second Law of thermodynamics Entropy changes in reversible and irreversible processes Variation of entropy with temperature pressure and volume Entropy concept as a measure of unavailable energy and criteria for the spontaneity of reaction Free energy enthalpy functions and their significance criteria for spontaneity of a process Partial molar quantities free energy volume heat concept Gibb s Duhem equation Chapter 3 Chemical Dynamics I Effect of temperature on reaction rates Rate law for opposing reactions of 1st order and 2nd order Rate law for consecutive Collision theory of reaction rates and its limitations Steric factor Activated complex theory Ionic reactions single and double sphere models Influence of solvent and ionic strength The comparison of collision and activated complex theory Chapter 4 Electrochemistry I Ion Ion Interactions The Debye Huckel theory of ion ion interactions Potential and excess charge density as a function of distance from the central ion Debye Huckel reciprocal length Ionic cloud and its contribution to the total potential Debye Huckel limiting law of activity coefficients and its limitations Ion size effect on potential Ion size parameter and the theoretical mean activity coefficient in the case of ionic clouds with finite sized ions Debye Huckel Onsager treatment for aqueous solutions and its limitations Debye Huckel Onsager theory for non aqueous solutions The solvent effect on the mobility at infinite dilution Equivalent conductivity vs concentration  $c^{1/2}$  as a function of the solvent Effect of ion association upon conductivity Debye Huckel Bjerrum equation Chapter 5 Quantum Mechanics II Schrodinger wave equation for a particle in a three dimensional box The concept of degeneracy among energy levels for a particle in three dimensional box Schrodinger wave equation for a linear harmonic oscillator Zero point energy of a particle possessing harmonic motion and its consequence Schrodinger wave equation for three dimensional Rigid rotator Energy of rigid rotator Space quantization Schrodinger wave equation for hydrogen atom separation of variable in polar spherical coordinates and its solution Principle azimuthal and

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**A Textbook of Physical Chemistry** A. S. Negi, S. C. Anand, 1985 Written primarily to meet the requirements of students at the undergraduate level this book aims for a self learning approach The fundamentals of physical chemistry have been explained with illustrations diagrams tables experimental techniques and solved problems

**Problems in Physical Chemistry** K. K. Sharma, 1981 **Concepts Of Physical Chemistry** A. Goel, 2006 This book has been written for the students of under graduate and postgraduate level of the various universities in India A special feature of the book is that the text has been illustrated with a large number of line diagrams and the data presented in the form of numerous tables for reference and comparison In the preparation of text standard works and review by renowned author have been freely consulted and the reference given chapter wise At the end of the book will be found useful by those who wish to make a more detailed study of the topics discussed Contents Physical Properties and Chemical Constitution Molecular Weight Determination of Macromolecules and Theory of Gases Kinetic Molecular Dynamical and Chemical Equilibrium Ionic Equilibrium Electrolytic Conductance and Electrolytic Transference Theory of Dilute Solutions **A Textbook Of Physical Chemistry, 4E** K. K. Sharma, 2009-11 **Advanced Physical Chemistry** Mehra Harish C, 1978

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