

SOME BASIC CONCEPT OF CHEMISTRY

FORMULA - SHEET

BY - JAY SIR (J*)

① $PV = nRT$

P = Pressure (atm)

V = Volume (Litre)

n = mole

R = gas constant (0.0821 atm L/mol K)

T = Temperature (Kelvin 'K')

② $d_{\text{gas}} = \frac{PM}{RT}$

M = Molar mass

Baki salt upar hai

③ $\text{Density} = \frac{\text{mass}}{\text{volume}}$

④ $\frac{F - 32}{9} = \frac{C}{5} = \frac{K - 273}{5}$

F = Degree Fahrenheit

C = Degree Centigrade

K = Degree Kelvin

⑤ $1\text{m}^3 = 1000\text{L}$
 $1\text{dm}^3 = 1\text{L}$
 $1\text{cm}^3 = 1\text{ml}$

⑥ $1\text{amu} = \frac{1}{12} \times \text{mass of one } ^{12}\text{C isotope}$

$1\text{amu} = 1.66 \times 10^{-24}\text{gm}$

$1\text{amu} = \frac{1}{N_A}\text{gm}$

⑦ $\text{mole} = \frac{\text{mass (gm)}}{\text{molar mass}}$

$\text{mass} = \text{mole} \times \text{molar mass}$

⑧ $\text{mole} = \frac{\text{no. of particles (atom/molecule)}}{6.022 \times 10^{23}}$

$\text{no. of particles} = \text{mol} \times N_A$

⑨ $\text{mole} = \frac{\text{volume of gas at NTP/STP}}{22.4}$

$V_{\text{gas}} = \text{mol} \times 22.4$

⑩ मोल मात्र

"Kuch bhi diya ho, sabse pehle mole nikalo fir jo puchhe, wo mole ki help se nikalo"

$\text{mole} = \frac{W}{M} = \frac{N}{N_A} = \frac{V_L}{22.4}$

Assessment Of Basic Chemistry Concepts Answer Sheet

**Lawrence Litwack, Linda Linc, Dolores
Bower**

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regulatory implications Provides a conceptual and relatively simple mathematical approach to fate and transport modeling yet complex derivations of most equations are given in appendices Integrates the use of numerous software packages pC pH EnviroLab Simulators Water Wastewater and Global Issues and Fate 2016 Contains numerous easy to understand examples and problems along with answers for most end of the chapter problems and simulators for answers to fate and transport questions Includes numerous companion laboratory experiments with EnviroLab Requiring just a basic knowledge of algebra and first year college chemistry to start Pollutant Fate and Transport in Environmental Multimedia is an excellent textbook for upper level undergraduate and graduate faculty and students studying environmental engineering and science

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