

Total No. of Questions :4]

SEAT No. :

P598

[Total No. of Pages : 3

[5215] - 419  
T.Y.B.Sc.  
CHEMISTRY  
CH-341 : Physical Chemistry  
(2013 Pattern) (Paper - I) (Semester - IV)

Time : 2 Hours]

[Max. Marks :40

Instructions to the candidates:

- 1) All questions are compulsory.
- 2) Figures to the right indicates full marks.
- 3) Neat diagrams must be drawn wherever necessary.
- 4) Use of calculator and logarithmic table is allowed.
- 5) Actual calculations must be shown while solving the problems.

Q1) Answer the following:

[10]

- a) Assuming the element exists as fcc lattice and the cell length is  $10.67 \text{ \AA}$ , determine the distance between 100 plane.
- b) Define De-Broglie wavelength.
- c)  $^{32}\text{P}$  has a half life of  $3.2 \times 10^4$  years, calculate its decay constant.
- d) Find Miller indices of a crystal planes which cuts the three axes at multiple unit distance  $\frac{1}{2}$ ,  $\frac{1}{3}$  and  $\infty$ .
- e) Write the cell reaction for the following cell  $\text{Pb}|\text{PbSO}_4|\text{H}_2\text{SO}_4(2\text{M})||\text{CuSO}_4(0.1\text{M})|\text{Cu}$ .
- f) Define average life.
- g) Define : EMF of a cell.
- h) Define centre of symmetry.
- i) If the EMF of the cell at  $20^\circ\text{C}$  and  $30^\circ\text{C}$  are 0.6973 and 0.6998 volt respectively, calculate its temperature coefficient.
- j) What is plateau voltage?

P.T.O.

# Tybsc Chemistry Question Paper

**Amar Nath Bhutani**



## **Tybsc Chemistry Question Paper:**

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**Chemistry 8080/9080 , 720 Concept Building Questions of Organic Chemistry** Ajnish Kumar Gupta, Bharti Gupta,2020-05-05 The guiding principle in writing this book was to create a set of Test papers for students A test paper that presents the material in a way that they learn to solve all the questions of Organic Chemistry in conceptual and sequential way In Test paper we mixed all our teaching experience of 15 years along with theoretical and experimental knowledge to generate a series of test paper for all students to reason their way to a solution rather than memorize a multitude of facts hoping they don't run out of memory This Test paper covers 24 papers with all type of questions which can give you a clear cut picture of subject that you must know before examination Each paper includes 10 Single Correct Question SCQ 7 Multiple Correct Question MCQ 5 Assertion Reason A R 2 Match the Column MTC 2 Comprehension 2 2 Questions 2 Integer i e 30 Questions in each paper Student can judge their preparation level by practicing in one hour These 24 papers are divided into 2 sets of 12 paper each In each set first of papers cover the whole organic chemistry into small segments Next 3 paper intermix two to three papers of the first eight paper and the last paper covers whole organic chemistry By doing this we want to revise your organic chemistry in 3 tiers so that not a single doubt should left out Organic chemistry is very easy and conceptual subject and need proper understanding of the basics and strategy to solve the questions in correct manner This Test paper will prepare your right mindset for learning Organic Chemistry This mindset is essentially the one that focuses you on a small number of straight forward fundamental concepts and helps you to apply them in different ways to solve the variety of problems you face in organic chemistry In this book balance has to be achieved between the number of questions and the quality of the questions especially because it is relatively easy to frame a very large number of multiple choice questions and theory of the subject The questions in this book have been selected keeping three things in mind First the questions are such that they really test the understanding of the subject Second the questions cover all concepts Third the number of questions has been kept large enough to offer meaningful practice to the students MCQs Chemistry V. K. Ahluwalia,2013-12-30 This book has been divided in 37 chapters under three parts Inorganic Chemistry Organic Chemistry and Physical Chemistry for convenient understanding It also includes solved model test papers of the previous three years of AIIMS CBSE PMT CPMT UP to enable students to develop the skills of problem solving and time management essential for any entrance examination In addition to providing answers to all the questions detailed explanatory notes to selected difficult questions have also been provided to justify the answer A separate section of Assertions and Reasons is also given at the end of each chapter Exhaustive Question Bank Explanatory Notes and Hints Assertions Reasons Includes Pre solved papers of five years Models Test Papers of AIIMS CBSE PMT CPMT McGraw-Hill's 500 Physical Chemistry Questions: Ace Your

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