

# Introduction to Chemistry

## Section 1.1 The Stories of Two Chemicals

In your textbook, read about the ozone layer.

The words of the boxes below just need to complete the passage.

atmosphere	oxygen gas	ozone	ozone hole
stratosphere	hydrophobic	ultraviolet radiation	

Earth's **OA** \_\_\_\_\_ is made up of several layers. The air molecules gather up near the lower level. This layer is called the **OB** \_\_\_\_\_. The next layer up is called the **OC** \_\_\_\_\_. This layer contains a protective \_\_\_\_\_.

**OD** \_\_\_\_\_ layer

Chlorine compounds **OE** \_\_\_\_\_, is made by ultraviolet radiation in the upper part of the atmosphere. The ozone layer's layer around Earth, which shields \_\_\_\_\_.

**OF** \_\_\_\_\_ Without ozone, you get more **OG** \_\_\_\_\_ to get a rash or possibly skin cancer. The thinning of the ozone layer, called the **OH** \_\_\_\_\_, is measured because without ozone all organisms on Earth are subject to more harmful \_\_\_\_\_.

In your textbook, read about chlorofluorocarbons.

The words statements below, rephrase as follows:

- \_\_\_\_\_ **I**. CFCs is another name for a chlorofluorocarbon.
- \_\_\_\_\_ **II**. CFCs are made up of carbon, fluorine, and chlorine.
- \_\_\_\_\_ **III**. All CFCs are synthetic chemicals.
- \_\_\_\_\_ **IV**. CFCs readily react easily with other chemicals.
- \_\_\_\_\_ **V**. CFCs were developed as replacements for toxic refrigerants.

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**S Ashworth**



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## **Table of Contents Chapter 11 Study For Content Mastery**

1. Understanding the eBook Chapter 11 Study For Content Mastery
  - The Rise of Digital Reading Chapter 11 Study For Content Mastery
  - Advantages of eBooks Over Traditional Books
2. Identifying Chapter 11 Study For Content Mastery
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Chapter 11 Study For Content Mastery
  - User-Friendly Interface
4. Exploring eBook Recommendations from Chapter 11 Study For Content Mastery
  - Personalized Recommendations
  - Chapter 11 Study For Content Mastery User Reviews and Ratings
  - Chapter 11 Study For Content Mastery and Bestseller Lists
5. Accessing Chapter 11 Study For Content Mastery Free and Paid eBooks

- Chapter 11 Study For Content Mastery Public Domain eBooks
  - Chapter 11 Study For Content Mastery eBook Subscription Services
  - Chapter 11 Study For Content Mastery Budget-Friendly Options
6. Navigating Chapter 11 Study For Content Mastery eBook Formats
    - ePub, PDF, MOBI, and More
    - Chapter 11 Study For Content Mastery Compatibility with Devices
    - Chapter 11 Study For Content Mastery Enhanced eBook Features
  7. Enhancing Your Reading Experience
    - Adjustable Fonts and Text Sizes of Chapter 11 Study For Content Mastery
    - Highlighting and Note-Taking Chapter 11 Study For Content Mastery
    - Interactive Elements Chapter 11 Study For Content Mastery
  8. Staying Engaged with Chapter 11 Study For Content Mastery
    - Joining Online Reading Communities
    - Participating in Virtual Book Clubs
    - Following Authors and Publishers Chapter 11 Study For Content Mastery
  9. Balancing eBooks and Physical Books Chapter 11 Study For Content Mastery
    - Benefits of a Digital Library
    - Creating a Diverse Reading Collection Chapter 11 Study For Content Mastery
  10. Overcoming Reading Challenges
    - Dealing with Digital Eye Strain
    - Minimizing Distractions
    - Managing Screen Time
  11. Cultivating a Reading Routine Chapter 11 Study For Content Mastery
    - Setting Reading Goals Chapter 11 Study For Content Mastery
    - Carving Out Dedicated Reading Time
  12. Sourcing Reliable Information of Chapter 11 Study For Content Mastery
    - Fact-Checking eBook Content of Chapter 11 Study For Content Mastery
    - Distinguishing Credible Sources
  13. Promoting Lifelong Learning
    - Utilizing eBooks for Skill Development

- Exploring Educational eBooks
14. Embracing eBook Trends
- Integration of Multimedia Elements
  - Interactive and Gamified eBooks

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