

Stoichiometry

Section 11.1 What is stoichiometry?

In your textbook, read about stoichiometry and the balanced equation.

For each statement below, write *true* or *false*.

- _____ 1. The study of the quantitative relationships between the amounts of reactants used and the amounts of products formed by a chemical reaction is called stoichiometry.
- _____ 2. Stoichiometry is based on the law of conservation of mass.
- _____ 3. In any chemical reaction, the mass of the products is less than the mass of the reactants.
- _____ 4. The coefficients in a chemical equation represent not only the number of individual particles but also the number of moles of particles.
- _____ 5. The mass of each reactant and product is related to its coefficient in the balanced chemical equation for the reaction by its molar mass.

Complete the table below, using information represented in the chemical equation for the combustion of methanol, an alcohol.

methanol + oxygen \rightarrow carbon dioxide + water



Substance	Molar Mass (g/mol)	Number of Molecules	Number of Moles (mol)	Mass (g)
6. Methanol	32.05			
7. Oxygen gas	32.00			
8. Carbon dioxide	44.01			
9. Water	18.02			

10. What are the reactants? _____
11. What are the products? _____
12. What is the total mass of the reactants? _____
13. What is the total mass of the products? _____
14. How do the total masses of the reactants and products compare? _____

Chapter 11 Study Stoichiometry

Richard H. Langley, John Moore



Chapter 11 Study Stoichiometry:

Thermodynamics, Gas Dynamics, and Combustion Henry Clyde Foust III, 2021-12-07 This textbook provides students studying thermodynamics for the first time with an accessible and readable primer on the subject The book is written in three parts Part I covers the fundamentals of thermodynamics Part II is on gas dynamics and Part III focuses on combustion Chapters are written clearly and concisely and include examples and problems to support the concepts outlined in the text The book begins with a discussion of the fundamentals of thermodynamics and includes a thorough analysis of engineering devices The book moves on to address applications in gas dynamics and combustion to include advanced topics such as two phase critical flow and blast theory Written for use in Introduction to Thermodynamics Advanced Thermodynamics and Introduction to Combustion courses this book uniquely covers thermodynamics gas dynamics and combustion in a clear and concise manner showing the integral connections at an advanced undergraduate or graduate student level **5 Steps to a 5 AP Chemistry, 2010-2011 Edition** John Moore, Richard H. Langley, 2010-01-01 A Perfect Plan for the Perfect Score We want you to succeed on your AP exam That s why we ve created this 5 step plan to help you study more effectively use your preparation time wisely and get your best score This easy to follow guide offers you a complete review of your AP course strategies to give you the edge on test day and plenty of practice with AP style test questions You ll sharpen your subject knowledge strengthen your thinking skills and build your test taking confidence with Full length practice exams modeled on the real test All the terms and concepts you need to know to get your best score Your choice of three customized study schedules so you can pick the one that meets your needs The 5 Step Plan helps you get the most out of your study time Step 1 Set Up Your Study Program Step 2 Determine Your Readiness Step 3 Develop the Strategies Step 4 Review the Knowledge Step 5 Build Your Confidence Topics include Reactions and Periodicity Stoichiometry Gases Thermodynamics Spectroscopy Light and Electrons Bonding Solids Liquids and Intermolecular Forces Solutions and Colligative Properties Kinetics Equilibrium Electrochemistry Nuclear Chemistry and Organic Chemistry Also includes AP Chemistry practice exams AP Advanced Placement Program and College Board are registered trademarks of the College Entrance Examination Board which was not involved in the production of and does not endorse this product **5 Steps to a 5 AP Chemistry, 2008-2009 Edition** John T. Moore, Richard H. Langley, 2008-01-04 A PERFECT PLAN FOR THE PERFECT SCORE We want you to succeed on your AP exam That s why we ve created this 5 step plan to help you study more effectively use your preparation time wisely and get your best score This easy to follow guide offers you a complete review of your AP course strategies to give you the edge on test day and plenty of practice with AP style test questions You ll sharpen your subject knowledge strengthen your thinking skills and build your test taking confidence with Full length practice exams modeled on the real test All the terms and concepts you need to know to get your best score Your choice of three customized study schedules so you can pick the one that meets your needs The 5 Step Plan helps you get the most out of your study time Step

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