

Diffusion and Osmosis: Lab Quiz Answers

Study the set of five beakers shown here to answer questions 1 – 3:



1. Which beaker(s) contain(s) a solution that is hypertonic to the bag?

1. Beaker 3
2. Beakers 2 and 4
3. Beakers 1, 2, and 5
4. Beaker 4
5. Beakers 3 and 4

The correct answer is c. Hypertonic means that the solute concentration in the beaker is higher than in the bag. This is true for beakers 1, 2, and 5. Beaker 3 is isotonic to the bag, and beaker 4 is hypotonic to the bag.

2. Which bag would you predict to show the least change in mass at the end of the experiment?

1. The bag in Beaker 1
2. The bag in Beaker 2
3. The bag in Beaker 3
4. The bag in Beaker 4
5. The bag in Beaker 5

The correct answer is c. Since the two solutions shown are isotonic, there will be no net gain or loss of water, and therefore a negligible change in mass.

3. Arrange the beakers in order of the mass of the bags inside them after the experiment has run for 30 minutes. List the bag that loses the most mass first.

1. 1, 2, 3, 4, 5
2. 1, 5, 2, 3, 4
3. 4, 3, 2, 5, 1
4. 3, 2, 1, 4, 5
5. 2, 1, 5, 3, 4

Diffusion And Osmosis Lab Questions Answers

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