

Heat Equation

Rearranging the specific heat expression gives the heat equation.

$$\begin{array}{l} \text{energy} \\ \text{heat} \end{array} = \text{specific heat} \times \text{mass} \times \text{change in temperature} \\ = \quad c \quad \times \quad m \quad \times \quad \Delta T$$

Units:

- Energy → Joules (J) or calories (cal)
- Mass → grams (g) or kilograms (kg)
- Temperature change → °C or °F or K
- Specific heat → cal/g•°C or J/g•°C

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