

CHAPTER 14 STUDY GUIDE

Mixtures and Solutions

Section 14.1 Heterogeneous and Homogeneous Mixtures

In your textbook, read about suspensions and colloids.

For each statement below, write *true* or *false*.

- false 1. A solution is a mixture containing particles that settle out of the mixture if left undisturbed.
- true 2. The most abundant substance in a colloid is the dispersion medium.
- false 3. A colloid can be separated by filtration.
- true 4. A solid emulsion consists of a liquid dispersed in a solid.
- true 5. Whipped cream is an example of a foam.
- false 6. In an aerosol, the dispersing medium is a liquid.
- true 7. Brownian motion results from the collisions of particles of the dispersion medium with the dispersed particles.
- true 8. Dispersed particles in a colloid do not tend to settle out because they have polar or charged atomic groups on their surfaces.
- false 9. Stirring an electrolyte into a colloid stabilizes the colloid.
- true 10. Colloids demonstrate the Tyndall effect.

The table below lists the characteristics of particles in colloids, solutions, and suspensions. Place a check in the column of each mixture whose particles have a particular characteristic.

Characteristics of Particles	Colloid	Solution	Suspension
11. Less than 1 nm in diameter		✓	✓
12. Between 1 nm and 1000 nm in diameter	✓	✓	✓
13. More than 1000 nm in diameter	✓	✓	
14. Settle out if undisturbed			
15. Pass through standard filter paper	✓		
16. Lower vapor pressure			✓
17. Scatter light	✓		

Chapter 14 Study Guide Chemistry Mixtures Solutions

David R. Klein



Chapter 14 Study Guide Chemistry Mixtures Solutions:

The Practice of Chemistry Study Guide & Solutions Manual Pamela Mills, Amina El-Ashmawy, 2003-04-14 Designed to help students understand the material better and avoid common mistakes Also includes solutions and explanations to odd numbered exercises *Organic Chemistry Study Guide with Solutions Manual* Neil E. Schore, 2007 The guide includes chapter introductions that highlight new material chapter outlines detailed comments for each chapter section a glossary and solutions to the end of chapter problems presented in a way that shows students how to reason their way to the answer

Chemistry, Student Study Guide James E. Brady, Fred Senese, 2008-01-28 The image on the front cover depicts a carbon nanotube emerging from a glowing plasma of hydrogen and carbon as it forms around particles of a metal catalyst Carbon nanotubes are a recently discovered allotrope of carbon Three other allotropes of carbon buckyballs graphite and diamond are illustrated at the left as is the molecule methane CH_4 from which nanotubes and buckyballs can be made The element carbon forms an amazing number of compounds with structures that follow from simple methane found in natural gas to the complex macromolecules that serve as the basis of life on our planet The study of chemistry also follows from the simple to the more complex and the strength of this text is that it enables students with varied backgrounds to proceed together to significant levels of achievement **Organic Chemistry, 4e Student Solution Manual and Study Guide**

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Chemistry Key Concise Review Guide Sixth Edition Darrell D. Ebbing, 1999 **Student Study Guide and Solutions Manual to Accompany General, Organic, and Biochemistry** Katherine J. Denniston, Denniston Katherine, 2006 Study Guide and Solutions Manual for McMurry's Organic Chemistry, Fifth Edition Susan McMurry, 2000 Provides answers and explanations to all in text and end of chapter exercises Also includes summaries of name reactions functional group synthesis and reactions lists of reagents and abbreviations and articles on topics ranging from infrared absorption frequencies to the Nobel Prize winners in Chemistry This edition now includes all new artwork expanded in text problems summary quizzes approximately every three chapters more detailed explanations in solutions and chapter outlines *Know Your 'O' Level Chemistry - A Study Guide*, **Study Guide and Solutions Manual for Organic Chemistry** Susan McMurry, 1996 This revision of the best selling organic chemistry textbook today has been fully updated and revised to offer more applications a

completely new chapter and dozens of new problems and examples McMurry's text is currently in use at hundreds of colleges and universities throughout the United States and Canada and is an international bestseller from the United Kingdom to the Pacific Rim In this edition McMurry continues to do what he does best focus on the important material of the course and explain it in a concise clear way

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Student Study Guide to accompany Chemistry Martin Silberberg, Libby Weberg, 2005-01-06

Study Guide and Solutions Manual for Seager/Slabaugh's Chemistry for Today Seager/Slabaugh, 2004 The fifth edition of the Study Guide and Student Solutions Manual has been updated to reflect all of the changes to the text This ancillary tests the student on the learning objectives in each chapter and provides answers to all of the even numbered end of chapter exercises New additional activities have been added to include a review of each section of the chapter and a section entitled Tying It All Together with a Laboratory Application

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