

9. (b) Bases soluble in water are called alkalis. Only the oxides of sodium, potassium, and calcium are soluble in water, so these form sodium hydroxide potassium hydroxide and calcium hydroxide. These are the strongest bases.

10. **Acids react with metals to liberate**
gas

- (a) Oxygen (b) Carbon dioxide
(c) Hydrogen (d) Nitrogen

10. (c) All acids react with metals to liberate hydrogen gas
Metal + Acid → Metal salt + Hydrogen gas

11. **Choose the correct statement(s)**

- (i) Most of the acids are water soluble
(ii) Acids react with metallic oxides and hydroxides to form metallic salt and water only.
(iii) Acids react with metallic carbonates to form metallic salt and hydrogen gas and water
(iv) Acetic acid is used as a food preservative
(a) (i) & (ii) only (b) (iii) & (iv)
(c) (i), (ii) & (iv) (d) all the above

11. (c) The third statement is wrong because acid reacts with metal carbonates to form metallic salt, carbon dioxide gas and water

12. **The reaction of acid and a base to form salt and water only is called**

- (a) Neutralization (b) Addition
(c) Substitution (d) Oxidation

12. (a) A neutralization reaction is the one in which acid and base react to form neutral salt and water.

13. **When ammonia gas is dissolved in water it reacts to form**

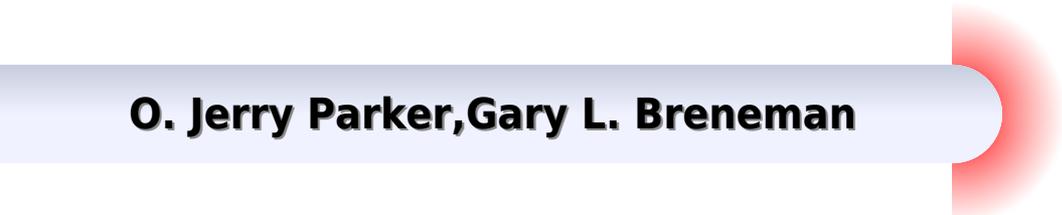
- (a) Ammonia water
(b) Ammonium hydroxide
(c) Ammonia acid
(d) All the above

13. (b) Ammonia reacts with water to form a hydroxide called ammonium hydroxide (NH₄OH)



Chapter 14 Acids And Bases Worksheet

O. Jerry Parker, Gary L. Breneman



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