

Stoichiometry Worksheet # 1

1) Glucose is used as a source of energy by the human body. The overall reaction in the body is



Calculate the number of moles of oxygen needed to oxidize 12.5 g of $\text{C}_6\text{H}_{12}\text{O}_6$.

2) Ammonia is synthesized from hydrogen and nitrogen according to the following equation.



If an excess of nitrogen is reacted with 3.41 moles of hydrogen gas, how many grams of ammonia (NH_3) can be produced?

3) Assume that in the decomposition of potassium chlorate, KClO_3 ,



80.5 moles of O_2 form. How many grams of potassium chlorite, the other product would be formed?

4) In a single displacement reaction,



9.23 moles of aluminum react with excess hydrochloric acid. How many moles of hydrogen will be produced?

5) The compound "Cisplatin" $\text{PtCl}_2(\text{NH}_3)_2$, has been found to be effective in treating some types of cancer. It can be synthesized using the following reaction



A. How many moles of "cisplatin" can be produced from 2.50 g K_2PtCl_4 ?

B. How many moles of NH_3 would be needed if 3.8 moles of K_2PtCl_4 were present?

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In some sort of inundated with monitors and the cacophony of quick transmission, the profound energy and emotional resonance of verbal art often diminish in to obscurity, eclipsed by the continuous barrage of noise and distractions. However, nestled within the lyrical pages of **Chapter 12 Stoichiometry Vocabulary Review Answers**, a fascinating perform of fictional beauty that pulses with fresh feelings, lies an memorable journey waiting to be embarked upon. Written by a virtuoso wordsmith, that mesmerizing opus instructions readers on a psychological odyssey, softly exposing the latent potential and profound impact stuck within the intricate web of language. Within the heart-wrenching expanse of the evocative examination, we will embark upon an introspective exploration of the book is central themes, dissect their captivating publishing style, and immerse ourselves in the indelible impact it leaves upon the depths of readers souls.

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