

14 • The Chemistry of Acids & Bases**STUDY QUESTIONS & PROBLEMS - KEY**

1. For the following aqueous equilibria, designate the Bronsted-Lowry conjugate acid-base pairs and establish the weaker side:



2. Write the name and formula for the conjugate bases of the following:



3. Complete the Bronsted-Lowry equilibria, label the components acid or base, and pair up the conjugate acid-base pairs:



4. Is the monohydrogen phosphate ion HPO_4^{2-} amphiprotic? -- YES

If so, write the formulas of its conjugate acid and its conjugate base. CA = H_2PO_4^- and CB = PO_4^{3-}

Chapter 17 Study Acids Bases

Clifford Lane



Chapter 17 Study Acids Bases:

Decoding **Chapter 17 Study Acids Bases**: Revealing the Captivating Potential of Verbal Expression

In an era characterized by interconnectedness and an insatiable thirst for knowledge, the captivating potential of verbal expression has emerged as a formidable force. Its ability to evoke sentiments, stimulate introspection, and incite profound transformations is genuinely awe-inspiring. Within the pages of "**Chapter 17 Study Acids Bases**," a mesmerizing literary creation penned with a celebrated wordsmith, readers embark on an enlightening odyssey, unraveling the intricate significance of language and its enduring effect on our lives. In this appraisal, we shall explore the book's central themes, evaluate its distinctive writing style, and gauge its pervasive influence on the hearts and minds of its readership.

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