

CBSE Class 12 Chemistry Sample Paper Set 8

CHEMISTRY
CLASS-XII

Time Allotted: 3 Hrs

Maximum Marks: 70

General Instructions:

1. All questions are compulsory.
2. Question No. 1-8 are very short answer questions and carry 1 mark each.
3. Question No. 9-18 are short answer questions and carry 2 marks each.
4. Question No. 19-27 are also short answer questions and carry 3 marks each.
5. Question No. 28-30 are long answer questions and carry 5 marks each.
6. Use log tables if necessary, use of calculators is not allowed.

Q.1 Bond enthalpy of fluorine is lower than that of chlorine, why?

Q.2 Give the IUPAC name of the following compound:



Q.3 On increasing temperature, activation energy of a reaction decreases, why?

Q.4 Which of the following is most effective electrolyte in the coagulative of $\text{AgI}/\text{Ag}+\text{sol}$?
 K_2SO_4 , MgCl_2 , $\text{K}_4[\text{Fe}(\text{CN})_6]$

Q.5 Write the reaction when glucose is heated with excess of HI.

Q.6 Which Xe compound has distorted octahedral shape?

Q.7 What is the denticity of co-ordination compound used for the treatment of lead Poisoning?

Q.8 An alkoxide is a stronger base than hydroxide ion. Justify.

Q.9 (a) State the law which helps to determine the limiting molar conductivity of weak electrolyte.

(b) Calculate limiting molar conductivity of CaSO_4 (limiting molar conductivity of calcium and sulphate ions are 119.0 and $160.0 \text{ S cm}^2 \text{ mol}^{-1}$ respectively).

Q.10 Rate constant K for first order reaction has been found to be $2.54 \times 10^{-3} \text{ sec}^{-1}$. Calculate its three-fourth life.

OR

A first order gas reaction $\text{A}_2(\text{g})\text{B}_2(\text{g}) \rightarrow 2\text{A}(\text{g}) + 2\text{B}(\text{g})$ at the temperature 400°C has the rate Constant $K = 2.0 \times 10^{-4} \text{ sec}^{-1}$. What percentage of A_2B_2 is decomposed on heating for 900 seconds.

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TABLE OF CONTENT

Fast Track Revision Stage 1 Question Solution Sample Question Paper 1 5 Stage 1 Questions Solutions Sample Question Paper 6 10 Stage 1 Unsolved Sample Question Paper 11 15 Latest CBSE Sample Paper One Day Before Exam

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Table of Contents Cbse 12th Sample Question Papers

1. Understanding the eBook Cbse 12th Sample Question Papers
 - The Rise of Digital Reading Cbse 12th Sample Question Papers
 - Advantages of eBooks Over Traditional Books
2. Identifying Cbse 12th Sample Question Papers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Cbse 12th Sample Question Papers
 - User-Friendly Interface
4. Exploring eBook Recommendations from Cbse 12th Sample Question Papers
 - Personalized Recommendations
 - Cbse 12th Sample Question Papers User Reviews and Ratings
 - Cbse 12th Sample Question Papers and Bestseller Lists
5. Accessing Cbse 12th Sample Question Papers Free and Paid eBooks
 - Cbse 12th Sample Question Papers Public Domain eBooks
 - Cbse 12th Sample Question Papers eBook Subscription Services
 - Cbse 12th Sample Question Papers Budget-Friendly Options
6. Navigating Cbse 12th Sample Question Papers eBook Formats
 - ePub, PDF, MOBI, and More
 - Cbse 12th Sample Question Papers Compatibility with Devices
 - Cbse 12th Sample Question Papers Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Cbse 12th Sample Question Papers
 - Highlighting and Note-Taking Cbse 12th Sample Question Papers
 - Interactive Elements Cbse 12th Sample Question Papers
8. Staying Engaged with Cbse 12th Sample Question Papers

- Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Cbse 12th Sample Question Papers
9. Balancing eBooks and Physical Books Cbse 12th Sample Question Papers
- Benefits of a Digital Library
 - Creating a Diverse Reading Collection Cbse 12th Sample Question Papers
10. Overcoming Reading Challenges
- Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Cbse 12th Sample Question Papers
- Setting Reading Goals Cbse 12th Sample Question Papers
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Cbse 12th Sample Question Papers
- Fact-Checking eBook Content of Cbse 12th Sample Question Papers
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
- Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
- Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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