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web solution average atomic mass of  $Li$  6.941 amu  $\times$  6.0151 amu  $\times$  7.0160 amu 6.941  $\times$  0.00925  $\times$  7.0160  $\times$  0.00925  $\times$  0.075  $\times$  0.075 corresponds to a natural abundance of 6%  $Li$  of 7.5 percent the natural abundance of  $Li$  is 0.925 or 92.5 percent 3.7 the unit factor required is 6.022

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web a molar mass  $Li_2CO_3$  2  $\times$  6.941 g 12.01 g 3  $\times$  16.00 g 73.89 g b molar mass  $CS_2$  12.01 g 2  $\times$  32.07 g 76.15 g c molar mass  $CHCl_3$

12 01 g 1 008 g 3 35 45 g 119 37 g d molar mass  $C_6H_8O_6$  6 12 01 g 8 1 008 g 6 16 00 g 176 12 g e molar mass  $KNO_3$  39 10 g  
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