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web polyatomic ions are charged groups of atoms an example is ammonium ion  $\text{NH}_4^+$  it has five atoms one nitrogen and four hydrogens that share a charge of 1 the polyatomic ions remain intact and parentheses may be required when using subscripts for example ammonium chloride is  $\text{NH}_4\text{Cl}$  and ammonium sulfide is  $\text{NH}_4\text{S}$

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there is more than one polyatomic ion then its formula is written