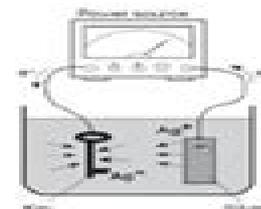


ENHANCING TEST ITEMS – CHEMISTRY (for Physical Sciences majors)

- Which of the following statements best describes the atomic radius of potassium atom and the atomic radius of a calcium atom?
 - The radius of the potassium atom is smaller because of its smaller nuclear charge.
 - The radius of the potassium atom is smaller because of its larger nuclear size.
 - The radius of the potassium atom is larger because its nuclear charge is smaller.
 - The radius of the potassium atom is larger because its nuclear charge is also larger.
- Which type of chemical reaction occurs when an iron nail rusts?
 - condensation
 - neutralization
 - oxidation-reduction
 - ionization-dissociation
- Which electron dot symbol represents the atom in period-4 with the highest first ionization energy?
 - 
 - 
 - 
 - 
- Which phase change is exothermic?
 - $\text{H}_2\text{O}_{(l)} \rightarrow \text{H}_2\text{O}_{(g)}$
 - $\text{H}_2\text{O}_{(g)} \rightarrow \text{H}_2\text{O}_{(l)}$
 - $\text{H}_2\text{O}_{(l)} \rightarrow \text{H}_2\text{O}_{(s)}$
 - $\text{H}_2\text{O}_{(s)} \rightarrow \text{H}_2\text{O}_{(l)}$
- Given the diagram, which statement BEST describes the key?
 - It is negative and acts as the cathode.
 - It is positive and acts as the cathode.
 - It is negative and acts as the anode.
 - It is positive and act as the anode
- Which technique is safest for diluting a concentrated acid with water?
 - Add the acid to the water quickly.
 - Add the water to the acid quickly.
 - Add the acid to the water slowly while stirring constantly.
 - Add the water to the acid slowly while stirring constantly.
- Which elements combine to form an ionic bond?
 - sodium and potassium
 - sodium and oxygen
 - carbon and oxygen
 - carbon and sulfur
- As the elements in group 16 are considered in order from top to bottom, the atomic radius of each successive element increases. This increase is primarily due to an increase in
 - electronegativity
 - ionization energy
 - the number of protons occupying the nucleus
 - the number of occupied principal energy levels
- An air pollution situation in which reactions occur on the surfaces of solid particles and produce acidic irritation of lung tissues is called _____.



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RM Cervero



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Table of Contents Chemistry Practice Test With Answers

1. Understanding the eBook Chemistry Practice Test With Answers
 - The Rise of Digital Reading Chemistry Practice Test With Answers
 - Advantages of eBooks Over Traditional Books
2. Identifying Chemistry Practice Test With Answers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Chemistry Practice Test With Answers
 - User-Friendly Interface
4. Exploring eBook Recommendations from Chemistry Practice Test With Answers
 - Personalized Recommendations
 - Chemistry Practice Test With Answers User Reviews and Ratings
 - Chemistry Practice Test With Answers and Bestseller Lists
5. Accessing Chemistry Practice Test With Answers Free and Paid eBooks
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 - Chemistry Practice Test With Answers Enhanced eBook Features
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 - Adjustable Fonts and Text Sizes of Chemistry Practice Test With Answers
 - Highlighting and Note-Taking Chemistry Practice Test With Answers
 - Interactive Elements Chemistry Practice Test With Answers

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 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Chemistry Practice Test With Answers
9. Balancing eBooks and Physical Books Chemistry Practice Test With Answers
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Chemistry Practice Test With Answers
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Chemistry Practice Test With Answers
 - Setting Reading Goals Chemistry Practice Test With Answers
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Chemistry Practice Test With Answers
 - Fact-Checking eBook Content of Chemistry Practice Test With Answers
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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