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Mole - Answer- SI unit for measuring the amount of substance;

Molar Mass - Answer- the mass of a mole of an element

Representative particles - Answer- refers to the species present in a substance: usually atoms, molecules, or formula units

Avogadro's number - Answer- the number of representative particles in a mole

Avogadro's hypothesis - Answer- states that equal volumes of gases at the same temperature and pressure contain equal numbers of particles

Molar volume - Answer- the quantity, 22.4 L

Dimensional analysis - Answer- mole to mass, mole to volume

Percent Composition - Answer- percent by mass of each element in the compound by which the relative amounts of the elements in a compound are expressed

STP (Standard Temperature and Pressure) - Answer- a temperature of 0°C and a pressure of 101.3 kPa, or 1 atmosphere (atm)

Empirical formula - Answer- the basic ratio gives the lowest whole-number ratio of the atoms of the elements contained in the compound

Molecular formula - Answer- either the same as its experimentally determined empirical formula, or it is a simple whole-number multiple of its empirical formula

Types of reactions - Answer- the five general types of reaction are combination, decomposition, single-replacement, double-replacement, and combustion

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J Dewey



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