

0620 Chemistry Past Papers (Year 2022) – Paper 2 MCQs

Feb/Mar 2022 – 22

1. Which gas has the fastest rate of diffusion?
- A  $\text{H}_2$                       B  $\text{CH}_4$                       C  $\text{CO}_2$                       D  $\text{SO}_2$
2. In which state does  $1 \text{ dm}^3$  of methane contain the most particles?
- A gas at  $100^\circ\text{C}$   
B gas at room temperature  
C liquid  
D solid
3. Which statement explains why isotopes of the same element have the same chemical properties?
- A They have the same electronic structure.  
B They have the same relative mass.  
C They have the same nucleon number.  
D They have the same proton number.

4. The electronic structures of atoms P and Q are shown.



P and Q form an ionic compound.

What is the formula of the compound?

- A PQ                      B  $\text{P}_2\text{Q}$                       C  $\text{P}_2\text{Q}_3$                       D  $\text{PQ}_3$
5. Fermentation of sugar produces a mixture of ethanol solution and solid yeast.
- How is the solid yeast removed from the mixture?
- A crystallisation  
B distillation  
C filtration  
D fractional distillation
6. Compound X contains carbon, hydrogen and oxygen only.
- By mass, it contains 26.7% carbon and 2.2% hydrogen.
- What is the empirical formula of X?

- A CHO                      B  $\text{C}_2\text{HO}$                       C  $\text{CH}_2\text{O}$                       D  $\text{CHO}_2$

# Chemistry 0620 Jun 2001

**Shasha Hu**





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