

UNIT 10: PERIODIC TABLE **CHAPTER 10: PERIODIC TABLE** **PERIODIC TABLE**

Section 10.1 (continued)

In your notebook, read aloud the modern periodic table.

Copy the letterheads in the boxes on the left below from the periodic table to complete the table on the right.

atomic mass	_____
atomic number	_____
electron configuration	_____
chemical name	_____
chemical symbol	_____

Put each item in Column A, under the letter of the matching item in Column B.

Column A

Column B

- | | |
|---|----------------------------|
| _____ 18. A column in the periodic table | a. metals |
| _____ 19. A row in the periodic table | b. group |
| _____ 20. Group 1 elements | c. period |
| _____ 21. Elements that are shiny and conduct electricity | d. representative elements |
| _____ 22. Group 17 elements | e. transition elements |

In the space on the left, write each of the statements in terms of the statement in bold. Change the underlined word or phrase to make it true.

- _____ 23. There are two representative elements of elements.
- _____ 24. Most elements besides boron of the elements in the periodic table are metalloids.
- _____ 25. Group 1A elements except for hydrogen are located on the left of metals.
- _____ 26. Group 2A elements are the alkaline earth metals.
- _____ 27. Group 7A elements are highly reactive nonmetals known as halogens.
- _____ 28. Group 8A elements are (a) nonmetals elements known as noble gases.
- _____ 29. Metalloids have properties of both metals and nonmetals.

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**Robert C. Smoot, Smoot, Richard G.
Smith, Jack Price**

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Table of Contents Chapter 5 Study Guide For Content Mastery Answer Key Chemistry

1. Understanding the eBook Chapter 5 Study Guide For Content Mastery Answer Key Chemistry
 - The Rise of Digital Reading Chapter 5 Study Guide For Content Mastery Answer Key Chemistry
 - Advantages of eBooks Over Traditional Books
2. Identifying Chapter 5 Study Guide For Content Mastery Answer Key Chemistry
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in a Chapter 5 Study Guide For Content Mastery Answer Key Chemistry
 - User-Friendly Interface
4. Exploring eBook Recommendations from Chapter 5 Study Guide For Content Mastery Answer Key Chemistry
 - Personalized Recommendations
 - Chapter 5 Study Guide For Content Mastery Answer Key Chemistry User Reviews and Ratings
 - Chapter 5 Study Guide For Content Mastery Answer Key Chemistry and Bestseller Lists

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5. Accessing Chapter 5 Study Guide For Content Mastery Answer Key Chemistry Free and Paid eBooks
 - Chapter 5 Study Guide For Content Mastery Answer Key Chemistry Public Domain eBooks
 - Chapter 5 Study Guide For Content Mastery Answer Key Chemistry eBook Subscription Services
 - Chapter 5 Study Guide For Content Mastery Answer Key Chemistry Budget-Friendly Options
6. Navigating Chapter 5 Study Guide For Content Mastery Answer Key Chemistry eBook Formats
 - ePub, PDF, MOBI, and More
 - Chapter 5 Study Guide For Content Mastery Answer Key Chemistry Compatibility with Devices
 - Chapter 5 Study Guide For Content Mastery Answer Key Chemistry Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Chapter 5 Study Guide For Content Mastery Answer Key Chemistry
 - Highlighting and Note-Taking Chapter 5 Study Guide For Content Mastery Answer Key Chemistry
 - Interactive Elements Chapter 5 Study Guide For Content Mastery Answer Key Chemistry
8. Staying Engaged with Chapter 5 Study Guide For Content Mastery Answer Key Chemistry
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Chapter 5 Study Guide For Content Mastery Answer Key Chemistry
9. Balancing eBooks and Physical Books Chapter 5 Study Guide For Content Mastery Answer Key Chemistry
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Chapter 5 Study Guide For Content Mastery Answer Key Chemistry
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Chapter 5 Study Guide For Content Mastery Answer Key Chemistry
 - Setting Reading Goals Chapter 5 Study Guide For Content Mastery Answer Key Chemistry
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Chapter 5 Study Guide For Content Mastery Answer Key Chemistry
 - Fact-Checking eBook Content of Chapter 5 Study Guide For Content Mastery Answer Key Chemistry
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
- Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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