

## Chemistry Final 2013 – Semester 1 Review Guide

### I. CLASSIFICATION OF MATTER

1. Define the following:

- a. Elements: \_\_\_\_\_
- b. Compound: \_\_\_\_\_
- c. Mixture: \_\_\_\_\_

2. What is the difference between homogeneous and heterogeneous mixtures?

3. Identify the following as an element (e), compound (c), or mixture (m):

Substance	E, C or M
As	
NaCl	
Sand + Water	
S	
Air	

4. Which of the following indicate a chemical change:

- (a) Fizzing  
(b) Burning  
(c) Color change  
(d) All of the above

5. Which of the following are physical changes:

- (a) State of changes (solid, liquid, gas)  
(b) Dissolving  
(c) Cracking open an egg  
(d) All of the above

6. What is the difference between a physical change and a chemical change?

### II. LABORATORY SAFETY

1. When should long hair be tied back?

2. When chemicals are spilled what should you do?

# Chemistry Midterm Exam Review

**Richard Bailey**



## **Chemistry Midterm Exam Review:**

## Reviewing **Chemistry Midterm Exam Review**: Unlocking the Spellbinding Force of Linguistics

In a fast-paced world fueled by information and interconnectivity, the spellbinding force of linguistics has acquired newfound prominence. Its capacity to evoke emotions, stimulate contemplation, and stimulate metamorphosis is really astonishing. Within the pages of "**Chemistry Midterm Exam Review**," an enthralling opus penned by a very acclaimed wordsmith, readers embark on an immersive expedition to unravel the intricate significance of language and its indelible imprint on our lives. Throughout this assessment, we shall delve to the book is central motifs, appraise its distinctive narrative style, and gauge its overarching influence on the minds of its readers.

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