

## GCSE Chemistry (AQA Higher Tier) Multiple Choice Questions

### Unit: C1

| Q# | Question   | A          | B          | C          | D          |
|----|--|------------|------------|------------|------------|
| 1  | Which of these is the correct group number for Magnesium?  | 1          | 2          | 3          | 4          |
| 2  | Which of these elements forms a 1+ ion when bonded ionically?  | Mg         | Na         | Pb         | Al         |
| 3  | Which of these elements is in Group 3?   | Al         | Pb         | Fe         | Br         |
| 4  | Which of these is the correct periodic table symbol for Gold?  | Gu         | Go         | Au         | Ag         |
| 5  | Which ion is formed by Aluminium when it bonds ionically?  | 1+         | 2+         | 3+         | 4+         |
| 6  | Which of these elements is liquid at room temperature?   | Al         | Fe         | He         | Br         |
| 7  | Which group contains a series of unreactive gases?   | 4          | 7          | 8/0        | 3          |
| 8  | Which of these elements, despite being in Group 1, does not have all the qualities of a metal?       | He         | H          | Hg         | Te         |
| 9  | Which of these is the correct mass and charge of a neutron?  | M:1<br>C:0 | M:1<br>C+1 | M:0<br>C:1 | M:0<br>C+1 |
| 10 | Three neutrons, two protons and one electron would have a cumulative charge equal to which of these? | +1         | -1         | +2         | -2         |
| 11 | Which of these is the correct approximate mass of an electron?                                       | 1/1200     | 1/1800     | 1/290      | 1/10       |

# Chemistry Gcse Questions And Answers

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